

## DESCRIPTION

### NOVEL COMPOUND, AND ELECTROPHOTOGRAPHIC PHOTORECEPTOR AND ELECTROPHOTOGRAPHIC APPARATUS USING THE SAME

#### TECHNICAL FIELD

[0001]

The present invention relates to a novel compound and in detail, it relates to a novel compound which is useful as a charge transport substance in an electrophotographic photoreceptor (hereinafter also referred to simply as "photoreceptor") and the like. Also, the invention relates to an electrophotographic photoreceptor and an electrophotographic apparatus and in detail, it relates to an electrophotographic photoreceptor comprising an electrically conductive substrate having thereon a photosensitive layer containing an organic material, which is used in printers or copiers of the electrophotographic system and the like, and to an electrophotographic apparatus.

#### BACKGROUND ART

[0002]

Hitherto, as a photosensitive layer of an electrophotographic photoreceptor, an inorganic photoconductive substance such as selenium and selenium alloys or a dispersions of an inorganic photoconductive substance such as zinc oxide and cadmium sulfide in a resin binding agent has been employed. In recent years, studies regarding electrophotographic photoreceptors using an organic photoconductive substance have advanced, some of which have been put into practical use while improving sensitivity and durability.

[0003]

Furthermore, the photoreceptor is required to have a function for holding a surface charge in a dark place, a function for receiving light to generate a charge, and a function for similarly receiving light to transport a charge. There are a so-called single layer type photoreceptor in which a single layer has these functions at the same time; and a so-called a layered type photoreceptor in which layers which have been functionally separated into a layer for principally contributing to the generation of a charge and a layer for contributing to holding of a surface charge in a dark place and charge transportation at the time of receiving light are laminated.

[0004]

For the image formation by the electrophotography using such a photoreceptor, for example, the Carlson system is applied. The image formation according to this system is carried out by electrically charging a photoreceptor in a dark place by corona discharge, forming an electrostatic latent image of letters, a picture, etc. of an original on the surface of the electrically charged photoreceptor, developing the formed electrostatic latent image with a toner, and fixing the developed toner image onto a support such as paper. The photoreceptor after transferring the toner image is subjected to neutralization, removal of a residual toner, optical neutralization, and the like and then provided for reuse.

[0005]

In comparison with inorganic photoreceptors, organic photoreceptors which are put into practical use are advantageous in view of flexibility, film forming properties, low costs, safety, etc., and due to the diversity of materials, further improvements in sensitivity, durability, and the like are being advanced.

[0006]

The most of organic photoreceptors are a layered type organic photoreceptor

in which the functions are separated into a charge generation layer and a charge transport layer. In general, in the layered type organic photoreceptor, a charge generation layer containing a charge generation substance such as pigments and dyes and a charge transport layer containing a charge transport substance such as hydrazine and triphenylamine are formed in this order on an electrically conductive support. Due to the properties of the charge transport substance which is electron donating, this layered type organic photoreceptor is of a hole transfer type and, when the surface of the photoreceptor is negatively charged, has sensitivity. Now, in a negative charge type, the corona discharge to be applied at the time of charging becomes unstable as compared with a positive charge type. Also, since ozone and nitrogen oxides are generated, they are adsorbed on the surface of the photoreceptor so that physical or chemical deterioration is liable to be caused and that a problem of worsening the environment is brought. From these viewpoints, in comparison with the negative charge type photoreceptor, the positive charge type photoreceptor having a large degree of freedom in the use condition is broad in its application range and advantageous as the photoreceptor.

[0007]

Then, in order to use a photoreceptor in the positive charge type, there is proposed a method for simultaneously dispersing a charge generation substance and a charge transport substance in a resin binder to use the photoreceptor in the form of a single photosensitive layer, and a part thereof is put into practical use. However, the single layer type photoreceptor is not sufficient with respect to the sensitivity for applying it in a high-speed machine, and further improvements are necessary in view of repeatability and the like.

[0008]

Furthermore, in order to take a layered structure of a function separation type

for the purpose of achieving high sensitivity, a method for laminating a charge generation layer on a charge transport layer to form a photoreceptor and using the photoreceptor in the positive charge type may be considered. However, according to this system, since the charge generation layer is formed on the surface, a problem is caused in the stability at the time of repeated use, etc. by corona discharge, light irradiation, mechanical abrasion, etc. In this case, it is also proposed to further provide a protective layer on the charge generation layer. However, though the mechanical abrasion is improved, a problem of, for example, bringing a reduction of electrical characteristics such as sensitivity is not overcome yet.

[0009]

In addition, there is also proposed a method for laminating an electron transporting charge transport layer on a charge generation layer to form a photoreceptor.

[0010]

As the electron transport substance, for example, 2,4,7-trinitro-9-fluorenone or the like is known. However, this substance is problematical in the safety because it is carcinogenic. Besides, in Patent Documents 1 to 4 and so on, cyano compounds and quinone based compounds are proposed. However, ones having electron transport ability sufficient for practical use have not been obtained yet.

[0011]

In addition, for example, in Patent Documents 5 to 15, Non-Patent Documents 1 to 4, and so on, a number of electron transport substances and electrophotographic photoreceptors using the same are proposed and described, and get attention. Furthermore, for example, as described in Patent Documents 16 to 20, photoreceptors using a combination of a hole transport substance and an electron transport substance in a single layer type photosensitive layer get attention for the reason that they have

high sensitivity, and a part thereof is put into practical use.

[0012]

On the other hand, up to date, the present inventors have proposed a variety of photoreceptors containing a substance having excellent electron transport properties (for example, Patent Documents 21 to 25 and so on).

[0013]

Patent Document 1: JP-A-1-206349

Patent Document 2: JP-A-6-59483

Patent Document 3: JP-A-9-190002

Patent Document 4: JP-A-9-190003

Patent Document 5: JP-A-4-360148

Patent Document 6: JP-A-3-290666

Patent Document 7: JP-A-5-92936

Patent Document 8: JP-A-9-151157

Patent Document 9: JP-A-5-279582

Patent Document 10: JP-A-7-179775

Patent Document 11: JP-A-10-73937

Patent Document 12: JP-A-4-338760

Patent Document 13: JP-A-1-230054

Patent Document 14: JP-A-8-278643

Patent Document 15: JP-A-2001-222122

Patent Document 16: JP-A-5-150481

Patent Document 17: JP-A-6-130688

Patent Document 18: JP-A-9-281728

Patent Document 19: JP-A-9-281729

Patent Document 20: JP-A-10-239874

Patent Document 21: JP-A-2000-75520

Patent Document 22: JP-A-2000-199979

Patent Document 23: JP-A-2000-143607

Patent Document 24: JP-A-2001-142239

Patent Document 25: JP-A-2001-215742

Non-Patent Document 1: *The Journal of Imaging Society of Japan*, Vol. 30,  
pp.266-273 (1991)

Non-Patent Document 2: *Preprint of Pan-Pacific Imaging Conference/Japan Hardcopy '98*, July 15-17, 1998, JA HALL, Tokyo, Japan, pp.207-210

Non-Patent Document 3: *Collected Papers of Japan Hardcopy '97*, July 9, 10  
and 11, 1997, JA HALL (Otemachi, Tokyo), pp.21-24

Non-Patent Document 4: *Collected Papers of Japan Hardcopy '92*, July 6, 7  
and 8, 1992, JA HALL (Otemachi, Tokyo), pp.173-176

## DISCLOSURE OF THE INVENTION

### PROBLEMS THAT THE INVENTION IS TO SOLVE

[0014]

As described above, a variety of investigations with respect to a charge transport substance having electron transport properties have hitherto been made. However, in recent years, according to a requirement for photoreceptors with high sensitivity, it has been demanded to realize a photoreceptor with high performance by using a new charge transport substance having more excellent electron transport properties.

[0015]

Then, an object of the invention is to provide a compound having excellent electron transport ability which is useful for applications of an electrophotographic

photoreceptor and to provide a positive charge type electrophotographic photoreceptor with high sensitivity for copiers and printers by using such a novel organic material as a charge transport substance in a photosensitive layer and an electrophotographic apparatus using the same.

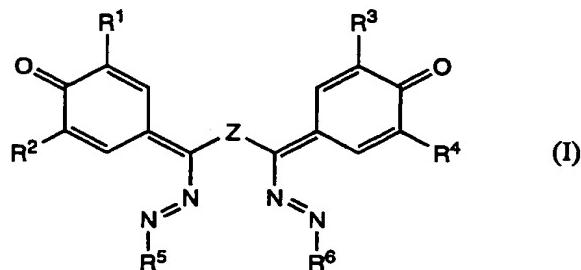
## MEANS FOR SOLVING THE PROBLEMS

[0016]

In order to achieve the foregoing object, the present inventors made extensive and intensive investigations regarding a variety of organic materials. As a result, it has been found that by using, as a charge transport substance, a specific compound having electron transport properties as represented by the following general formula (I), it is possible to obtain a photoreceptor with high sensitivity which can be used by positive charge, leading to accomplishment of the invention.

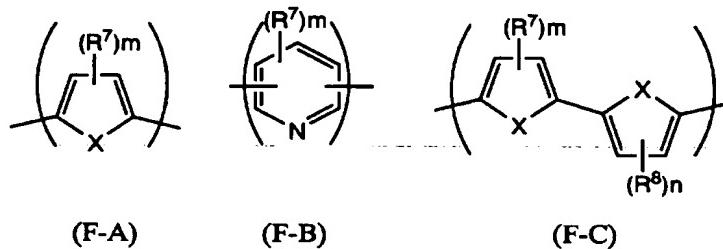
[0017]

Specifically, in order to solve the foregoing problems, the invention is concerned with a novel compound which is characterized by having a structure represented by the following general formula (I):



(in the general formula (I), R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, and R<sup>4</sup> may be the same or different and each represents a hydrogen atom, an optionally substituted alkyl group having from 1 to 12 carbon atoms, or an optionally substituted aryl group; R<sup>5</sup> and R<sup>6</sup> may be the same or different and each represents an optionally substituted aryl group or an optionally

substituted heterocyclic group; Z represents a structure represented by the following general formula (F-A), (F-B) or (F-C):



(in the formulae, R<sup>7</sup> and R<sup>8</sup> may be the same or different and each represents an optionally substituted alkyl group having from 1 to 12 carbon atoms; m and n each represents an integer of from 0 to 2; X represents a sulfur atom or an oxygen atom; and the substituents each represents a halogen atom, a nitro group, an alkyl group, an aryl group, a heterocyclic group, a halogenated alkyl group, or an alkoxy group, and the substituents may be taken together to form a ring); and the substituents each represents a halogen atom, a nitro group, an alkyl group, an aryl group, a heterocyclic group, a halogenated alkyl group, or an alkoxy group, and the substituents may be taken together to form a ring).

[0018]

Furthermore, in order to solve the foregoing problems, the invention is concerned with an electrophotographic photoreceptor comprising an electrically conductive substrate having thereon a photosensitive layer containing a charge generation substance and a charge transport substance, which is characterized in that the subject photosensitive layer contains at least one kind of the foregoing novel compound (I).

[0019]

In addition, the invention is concerned with an electrophotographic apparatus which is characterized by being provided with the foregoing electrophotographic

photoreceptor of the invention and performing a charge process by a positive charge process.

## EFFECT OF THE INVENTION

### [0020]

According to the invention, it is possible to obtain a compound having excellent electron transport properties, and by applying this compound to an electron device using an organic compound, such as electrophotographic photoreceptors, it becomes possible to enhance electrical characteristics and the like. Furthermore, according to the invention, since this compound is used as a charge transport substance in a photosensitive layer as provided on an electrically conductive substrate, it has become possible to obtain a photoreceptor with high sensitivity and excellent electrical characteristics in positive charge. Moreover, since it is possible to choose, as a charge generation substance, a substance which is suitable adaptive with the type of an exposure light source, by using a phthalocyanine compound, a squarylium compound, a bisazo compound, etc., it is possible to obtain a photoreceptor which can be used in semiconductor laser printers or copiers. In addition, if desired, it is possible to enhance the durability by providing a covering layer on the surface.

## BEST MODE FOR CARRYING OUT THE INVENTION

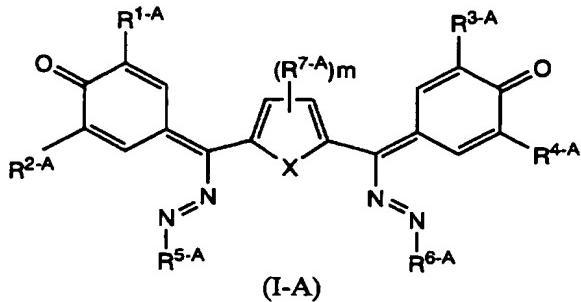
### [0021]

Preferred embodiments of the invention will be hereunder described in detail.

First of all, a compound represented by the foregoing general formula (I) will be described.

Examples of the compound represented by the foregoing general formula (I) include compounds represented by the following general formula (I-A) in which Z in

the general formula (I) is a structure represented by the foregoing general formula (F-A).

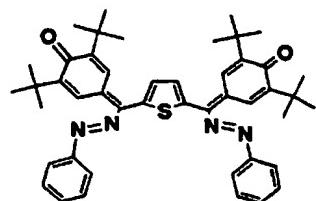


(In the formula (I-A),  $R^{1-A}$ ,  $R^{2-A}$ ,  $R^{3-A}$ , and  $R^{4-A}$  may be the same or different and each represents a hydrogen atom, an optionally substituted alkyl group having from 1 to 12 carbon atoms, or an optionally substituted aryl group;  $R^{5-A}$  and  $R^{6-A}$  may be the same or different and each represents an optionally substituted aryl group or an optionally substituted heterocyclic group;  $R^{7-A}$  represents an optionally substituted alkyl group having from 1 to 12 carbon atoms;  $X$  represents a sulfur atom or an oxygen atom;  $m$  represents an integer of from 0 to 2; and the substituents each represents a halogen atom, a nitro group, an alkyl group having from 1 to 6 carbon atoms, an aryl group, a halogenated alkyl group having from 1 to 6 carbon atoms, or an alkoxy group having from 1 to 6 carbon atoms.) Specific examples of compounds represented by the foregoing general formula (I-A) will be given by the following structural formulae (I-A-1) to (I-A-40), but it should not be construed that the invention is limited to these compounds. Incidentally, the term

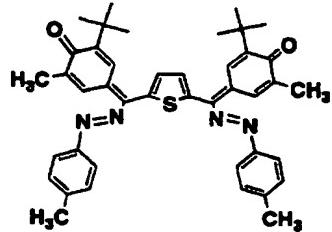


in the specific examples means a t-butyl group.

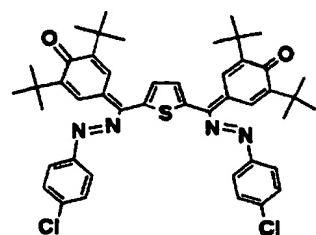
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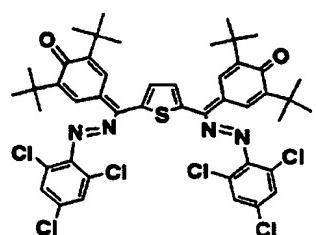
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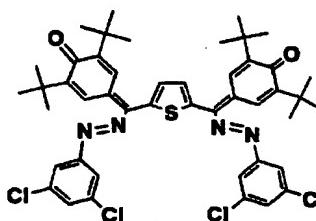
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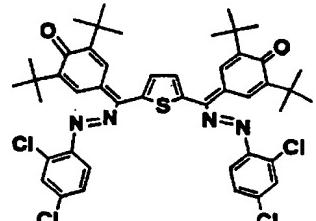
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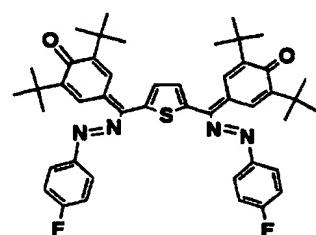
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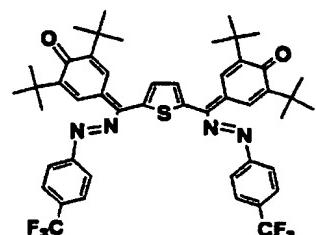
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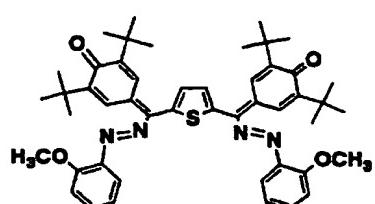
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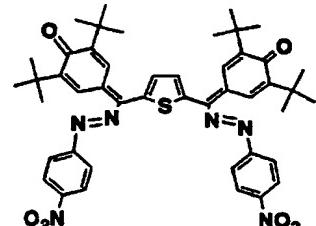
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(I-A-8)

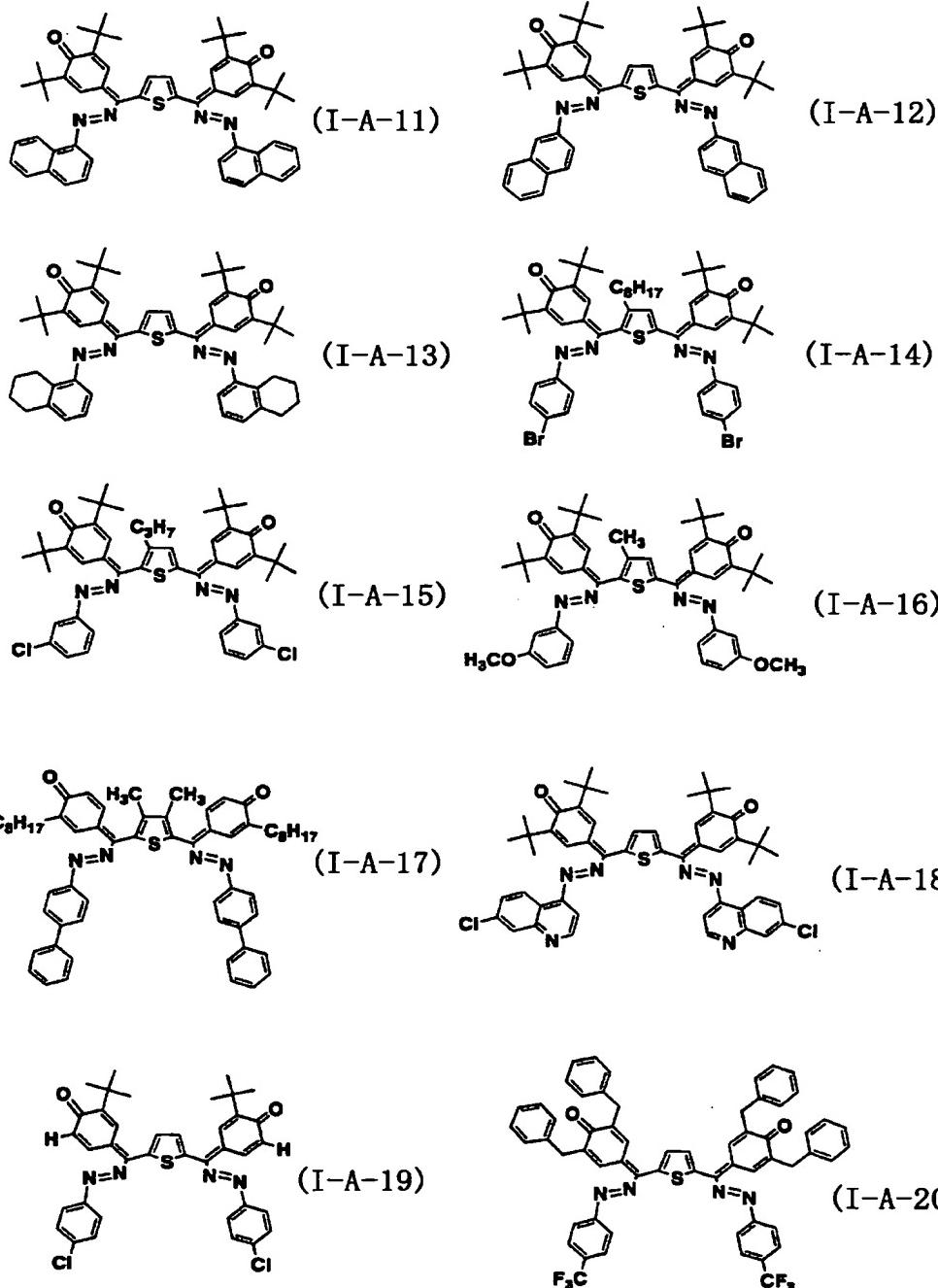


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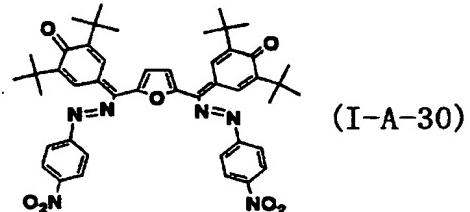
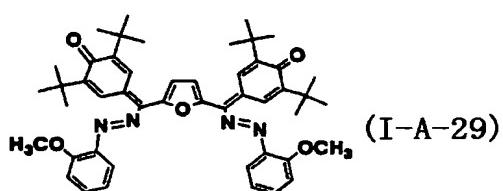
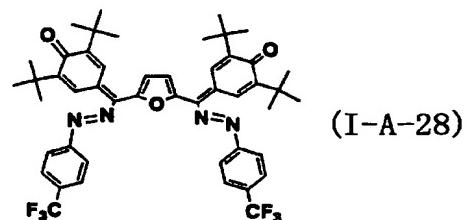
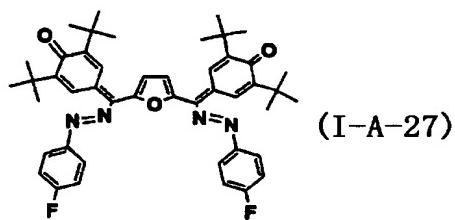
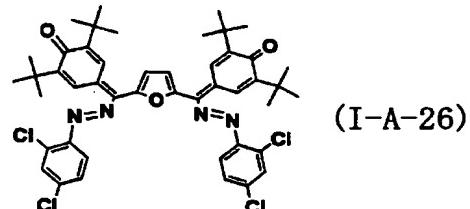
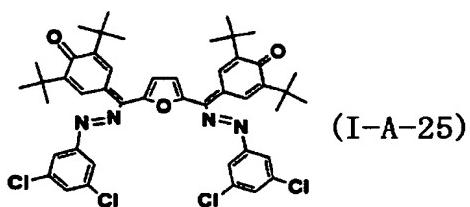
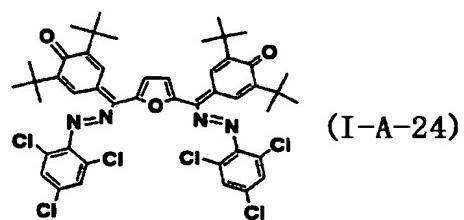
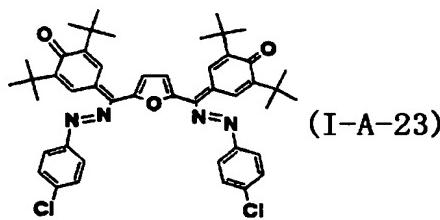
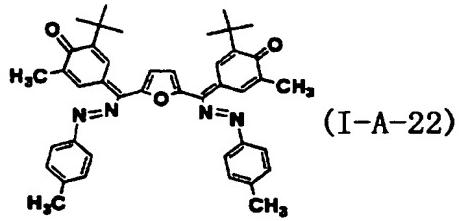
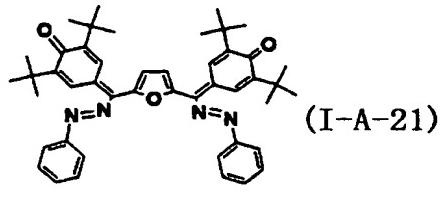


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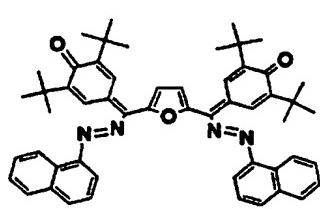
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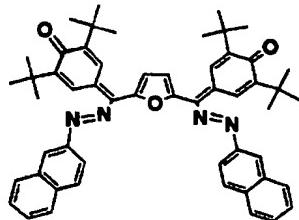
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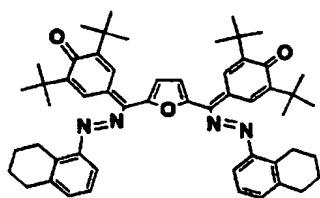
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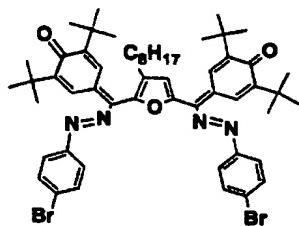
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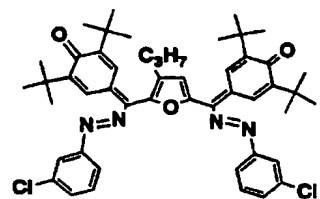
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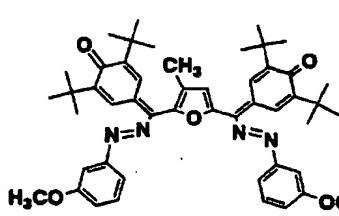
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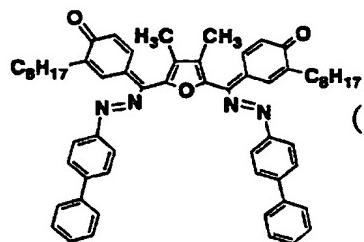
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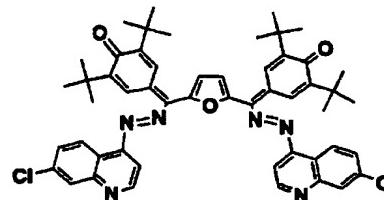
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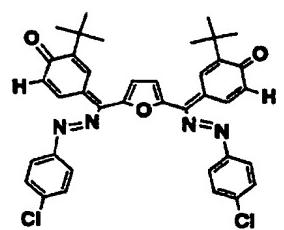
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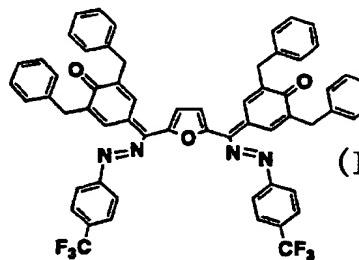
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(I-A-39)

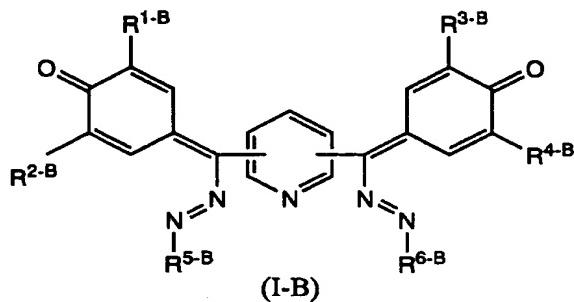


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[0026]

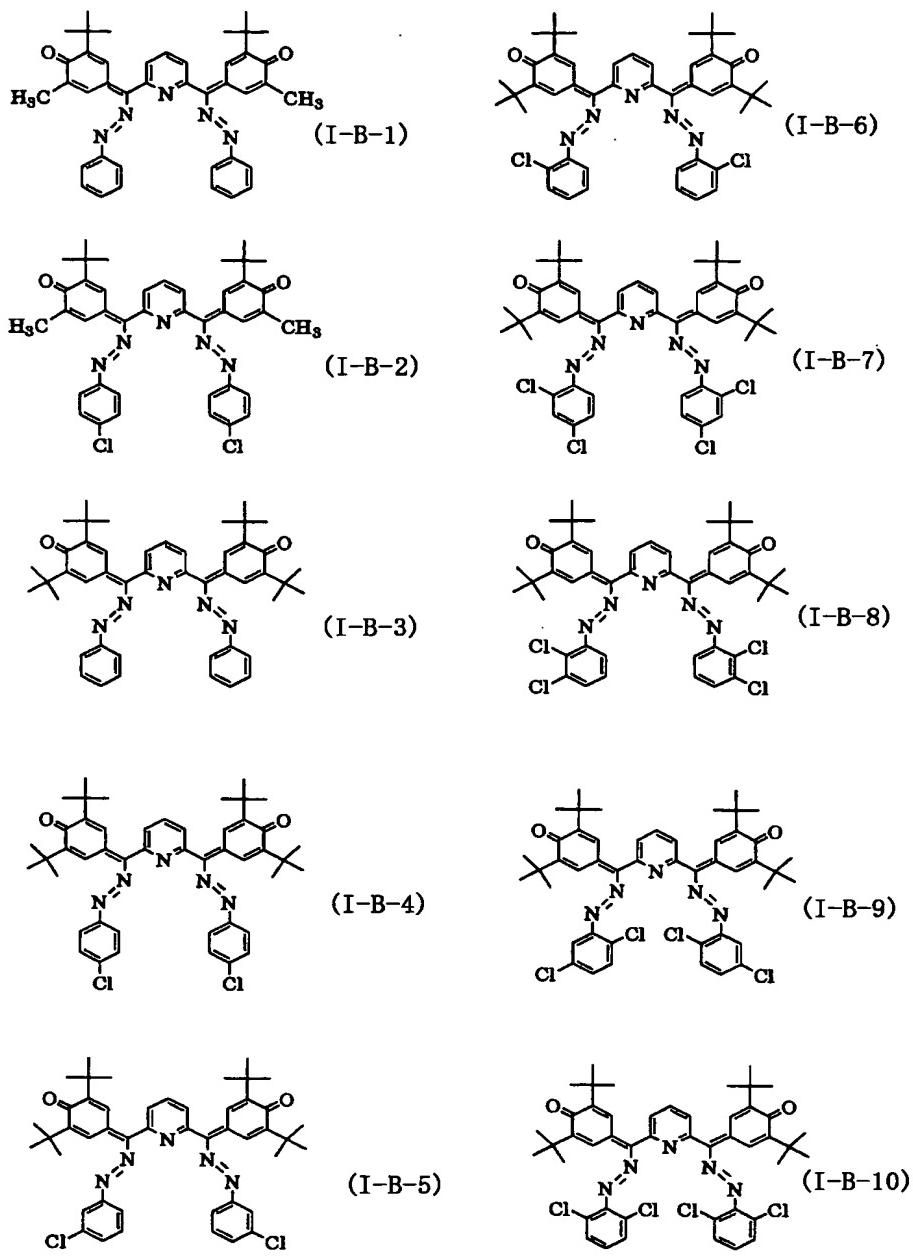
Furthermore, examples of the compound represented by the foregoing general

formula (I) include compounds represented by the following general formula (I-B) in which Z in the general formula (I) is a structure represented by the foregoing general formula (F-B) ( $m = 0$ ).

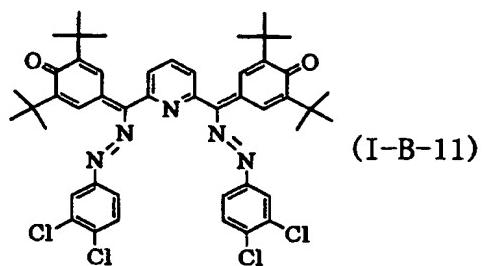


(In the formula (I-B),  $R^{1-B}$ ,  $R^{2-B}$ ,  $R^{3-B}$ , and  $R^{4-B}$  may be the same or different and each represents a hydrogen atom or an optionally substituted alkyl group having from 1 to 12 carbon atoms;  $R^{5-B}$  and  $R^{6-B}$  may be the same or different and each represents an optionally substituted aryl group or an optionally substituted heterocyclic group; and the substituents each represents a halogen atom, an alkyl group, an alkoxy group, an aryl group, a heterocyclic group, a fluorinated alkyl group, or a nitro group, and the substituents may be taken together to form a ring.) Specific examples of compounds represented by the foregoing general formula (I-B) will be given by the following structural formulae (I-B-1) to (I-B-30), and specific examples of other compounds represented by the foregoing general formula (I) will be given by the following structural formulae (I-B-31) to (I-B-40). However, it should not be construed that the invention is limited to these compounds.

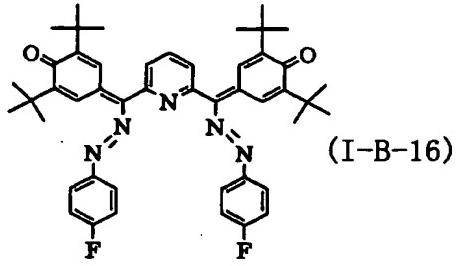
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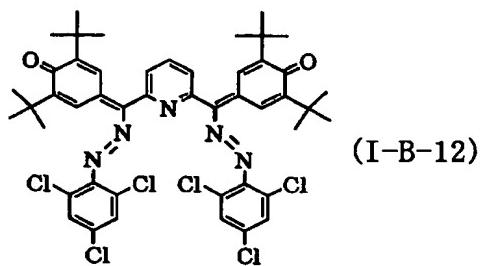
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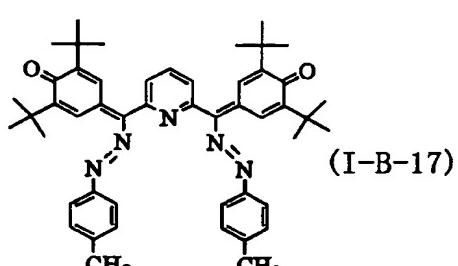
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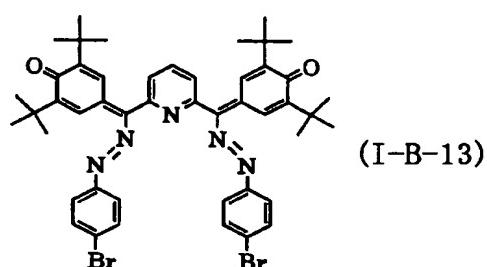
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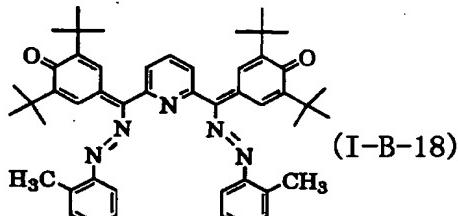
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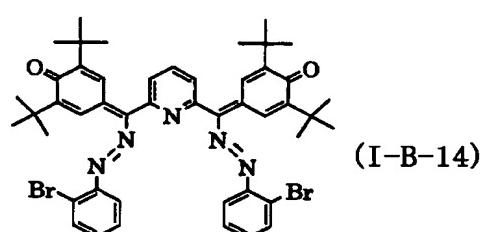
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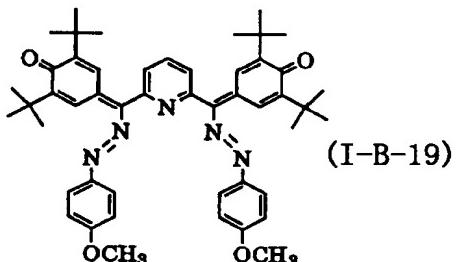
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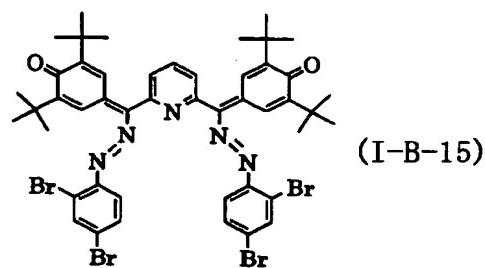
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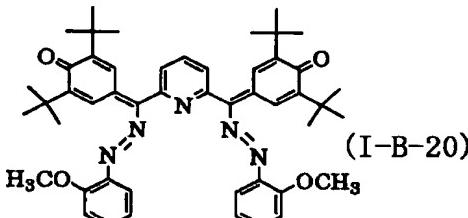
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(I-B-19)

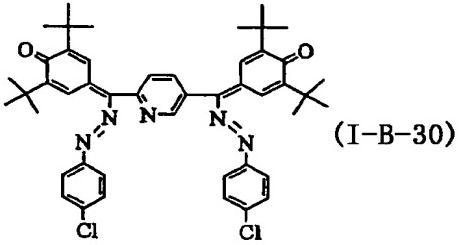
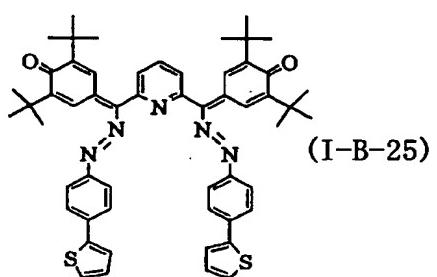
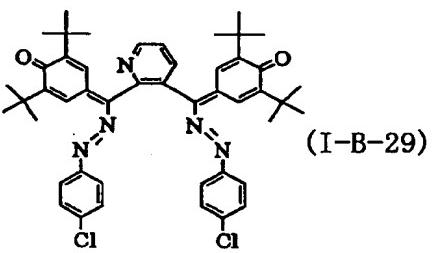
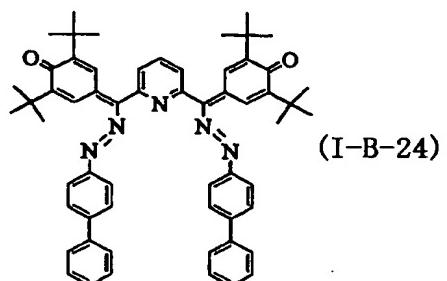
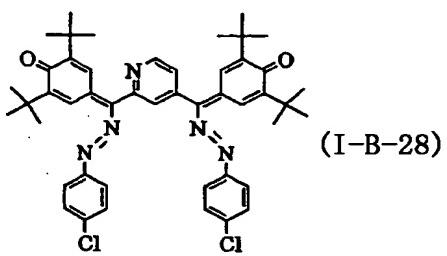
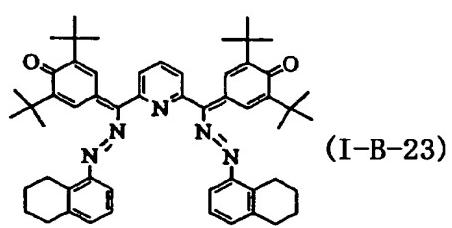
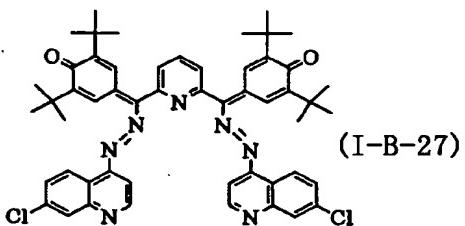
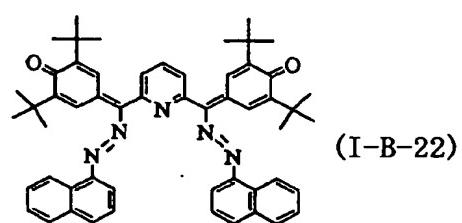
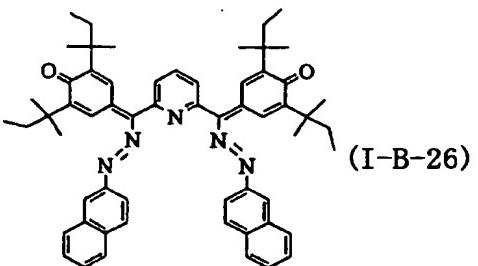
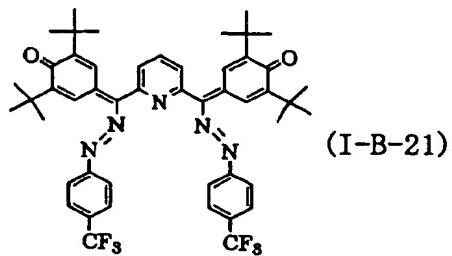


(I-B-15)

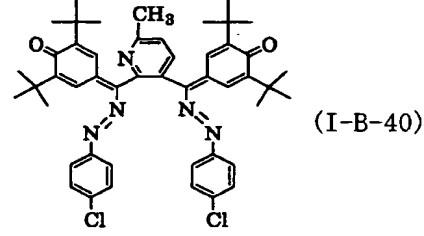
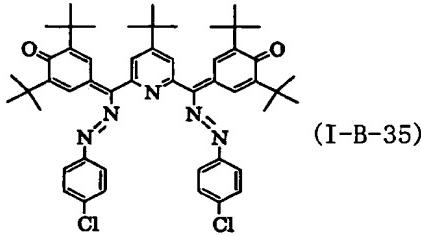
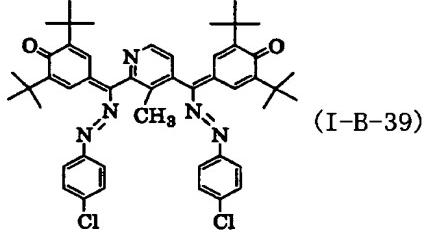
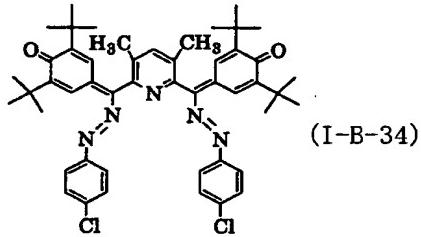
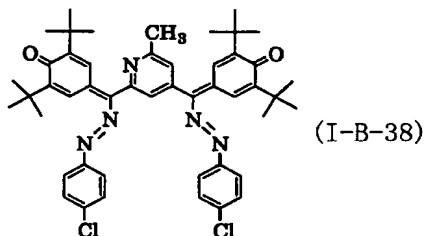
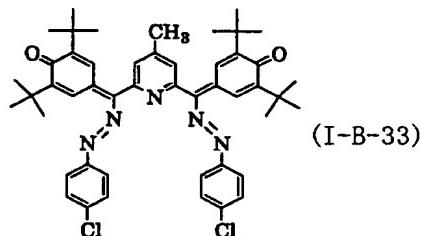
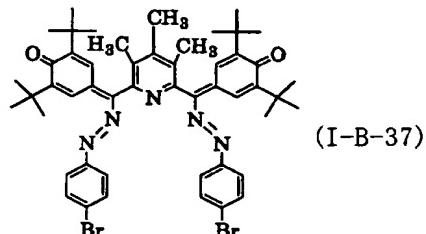
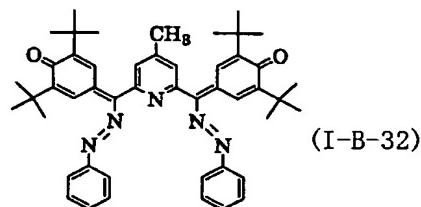
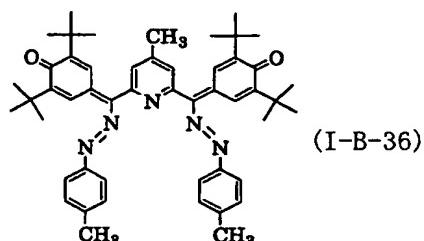
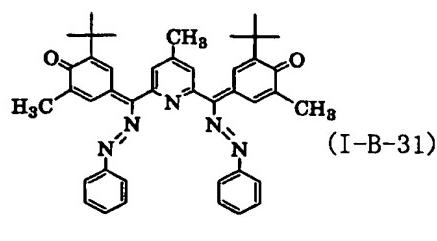


(I-B-20)

[0029]

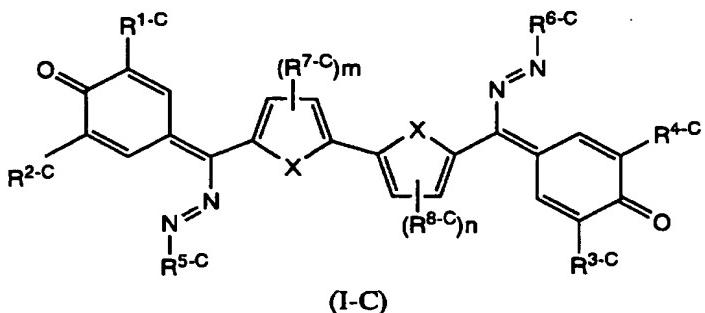


[0030]



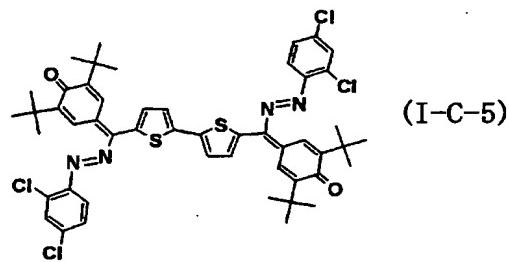
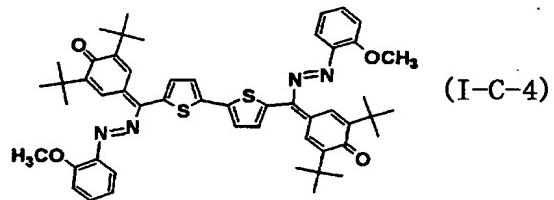
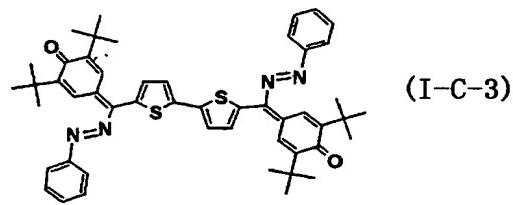
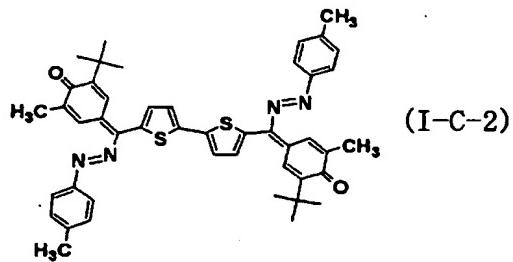
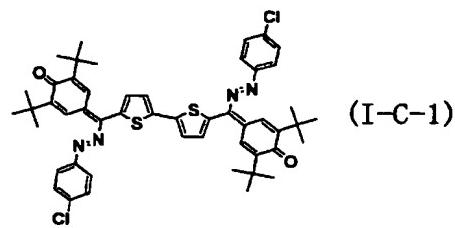
[0031]

In addition, examples of the compound represented by the foregoing general formula (I) include compounds represented by the following general formula (I-C) in which Z in the general formula (I) is a structure represented by the foregoing general formula (F-C).

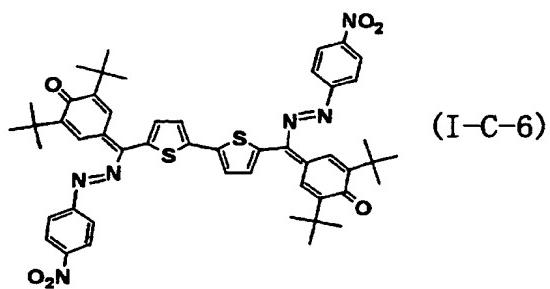


(In the formula (I-C),  $R^{1-C}$ ,  $R^{2-C}$ ,  $R^{3-C}$ , and  $R^{4-C}$  may be the same or different and each represents a hydrogen atom, an optionally substituted alkyl group having from 1 to 6 carbon atoms, or an optionally substituted aryl group;  $R^{5-C}$  and  $R^{6-C}$  may be the same or different and each represents an optionally substituted aryl group or a heterocyclic group;  $R^{7-C}$  and  $R^{8-C}$  each represents a hydrogen atom or an optionally substituted alkyl group having from 1 to 10 carbon atoms;  $X$  represents a sulfur atom or an oxygen atom;  $m$  and  $n$  each represents an integer of from 1 to 2; and the substituents each represents a halogen atom such as chlorine and bromine, a nitro group, an alkyl group having from 1 to 6 carbon atoms such as a methyl group, a cycloalkyl group, an aryl group, a heterocyclic group, a halogenated alkyl group, or an alkoxy group having from 1 to 6 carbon atoms such as a methoxy group, and the substituents may be taken together to form a ring.) Specific examples of compounds represented by the foregoing general formula (I-C) will be given by the following structural formulae (I-C-1) to (I-C-19), but it should not be construed that the invention is limited to these compounds.

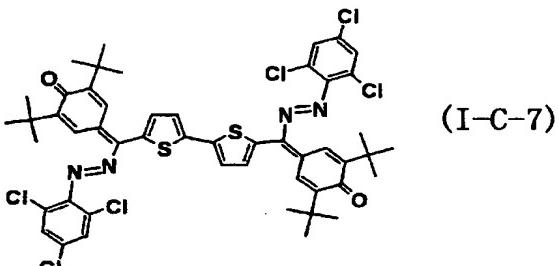
[0032]



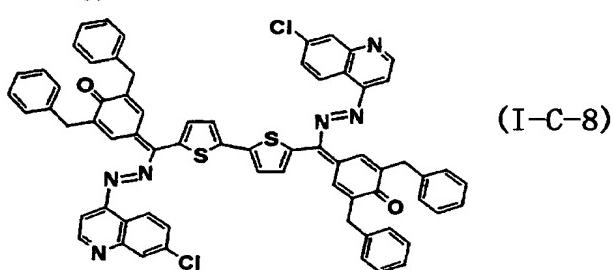
[0033]



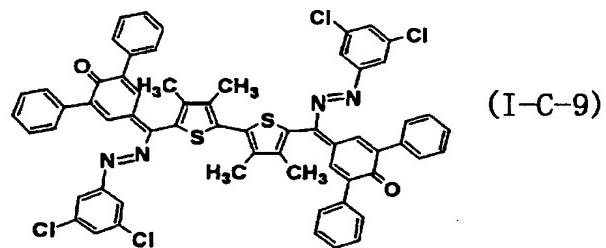
(I-C-6)



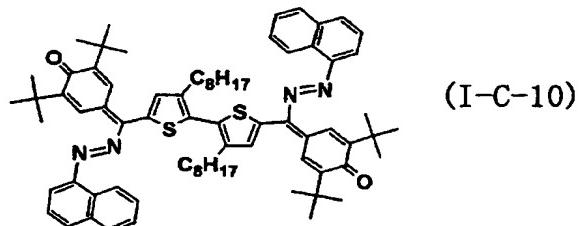
(I-C-7)



(I-C-8)

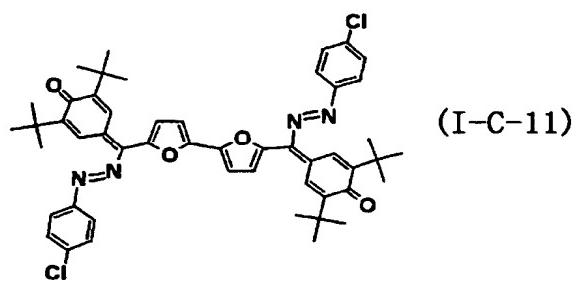


(I-C-9)

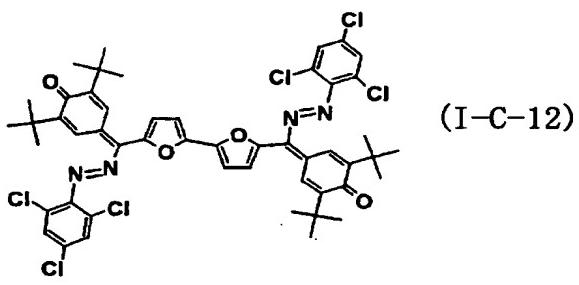


(I-C-10)

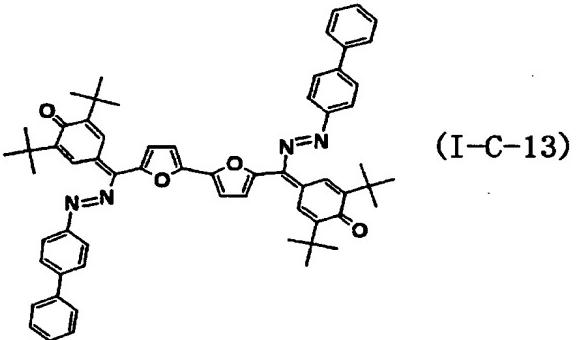
[0034]



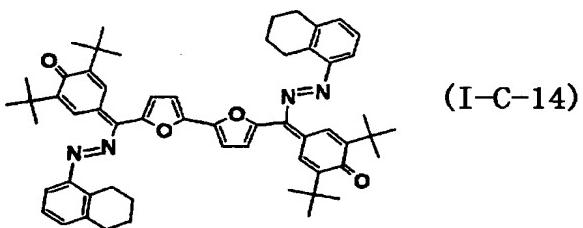
(I-C-11)



(I-C-12)

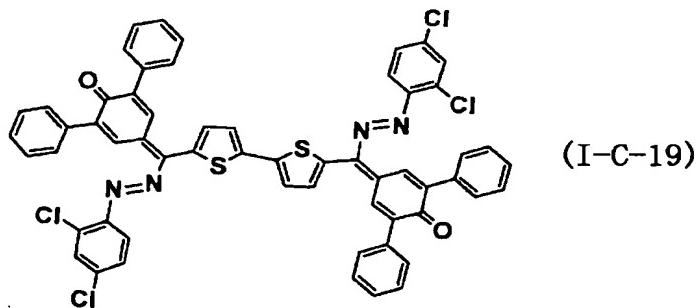
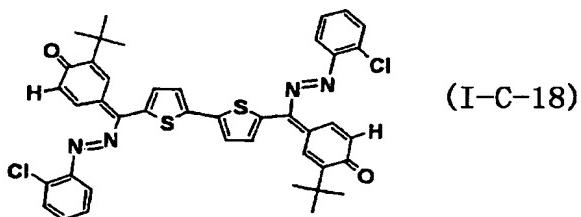
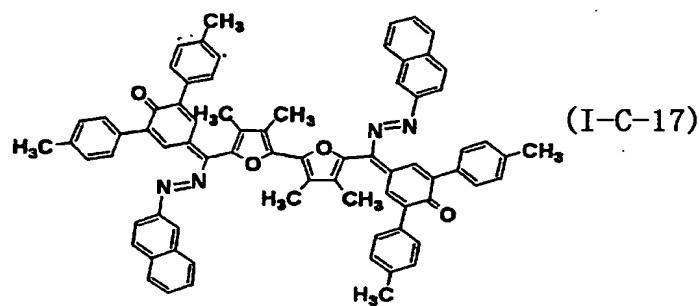
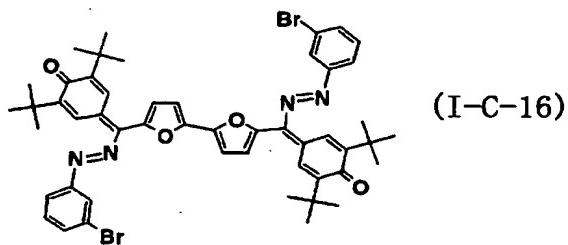
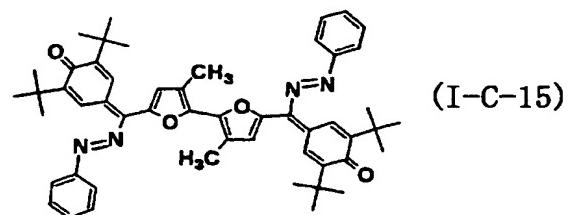


(I-C-13)



(I-C-14)

[0035]



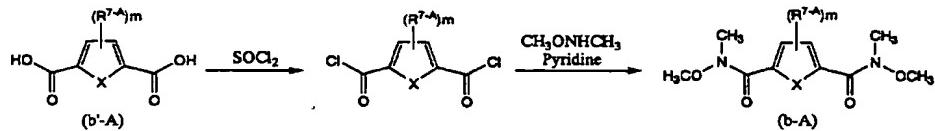
[0036]

The compound represented by the foregoing general formula (I-A) according

to invention can be, for example, synthesized according to the following reaction schemes (1-A) and (2-A). That is, first of all, as shown in the following reaction scheme (1-A), a compound represented by the structural formula (b'-A) is synthesized from a compound represented by the structural formula (b''-A). Next, as shown in the following reaction scheme (2-A), this compound represented by the structural formula (b'-A) is allowed to react with a compound represented by the structural formula (a-A) and/or a compound represented by the structural formula (a'-A) and an appropriate organometallic reagent (for example, magnesium), and thereafter, a protective group (TMS: trimethylsilyl group) is eliminated off to synthesize a compound represented by the structural formula (c-A). In addition, this compound represented by the structural formula (c-A) is subjected to dehydration condensation with a compound represented by the structural formula (d-A) and/or a compound represented by the structural formula (d'-A), followed by oxidation with an appropriate catalyst (for example, lead dioxide ( $PbO_2$ )). Thus, the compound represented by the structural formula (I-A) can be synthesized. Incidentally, the term "TBAF" in the following reaction scheme (2-A) means tetrabutylammonium fluoride.

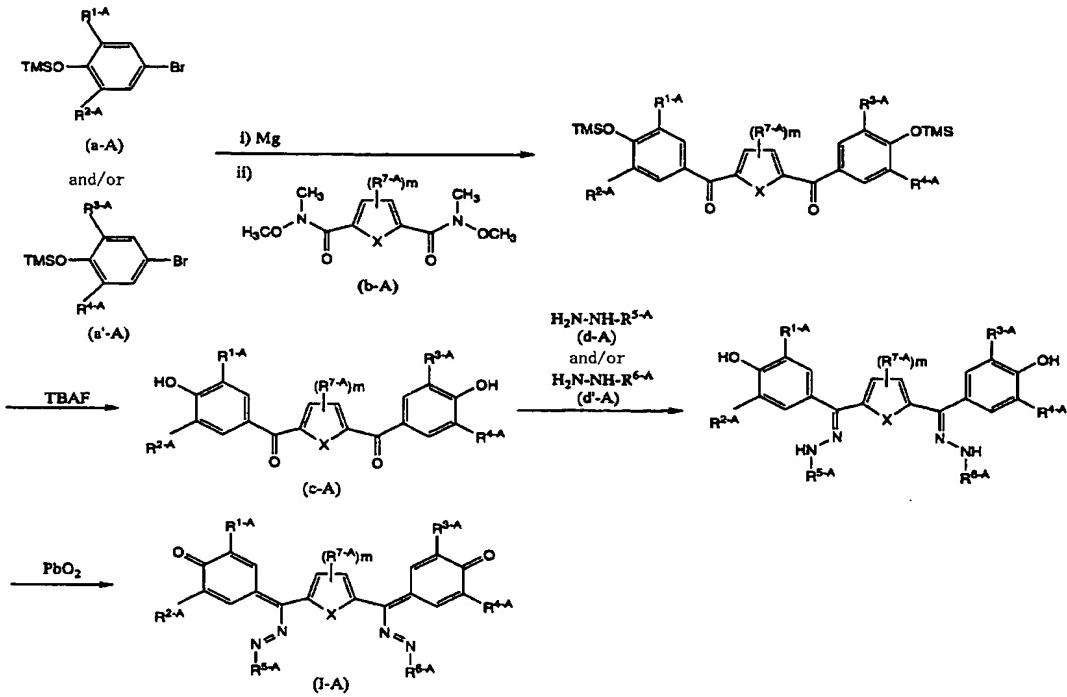
[0037]

Reaction scheme (1-A)



[0038]

Reaction scheme (2-A)



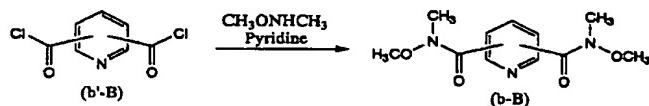
[0039]

Furthermore, the compound represented by the foregoing general formula (I-B) according to invention can be, for example, synthesized according to the following reaction schemes (1-B) and (2-B). That is, first of all, as shown in the following reaction scheme (1-B), a compound represented by the structural formula (b-B) is synthesized from a compound represented by the structural formula (b'-B). Next, as shown in the following reaction scheme (2-B), the compound represented by the structural formula (b-B) is allowed to react with a compound represented by the structural formula (a-B) and/or a compound represented by the structural formula (a'-B) and an appropriate organometallic reagent (for example, magnesium), and thereafter, a protective group (TMS: trimethylsilyl group) is eliminated off to synthesize a compound represented by the structural formula (c-B), followed by dehydration condensation with a compound represented by the structural formula (d-B) and/or a compound represented by the structural formula (d'-B) and subsequent

oxidation with an appropriate catalyst (for example, lead dioxide ( $PbO_2$ )). Thus, the compound represented by the structural formula (I-B) can be synthesized. Incidentally, the term "TBAF" in the following reaction scheme (2-B) means tetrabutylammonium fluoride.

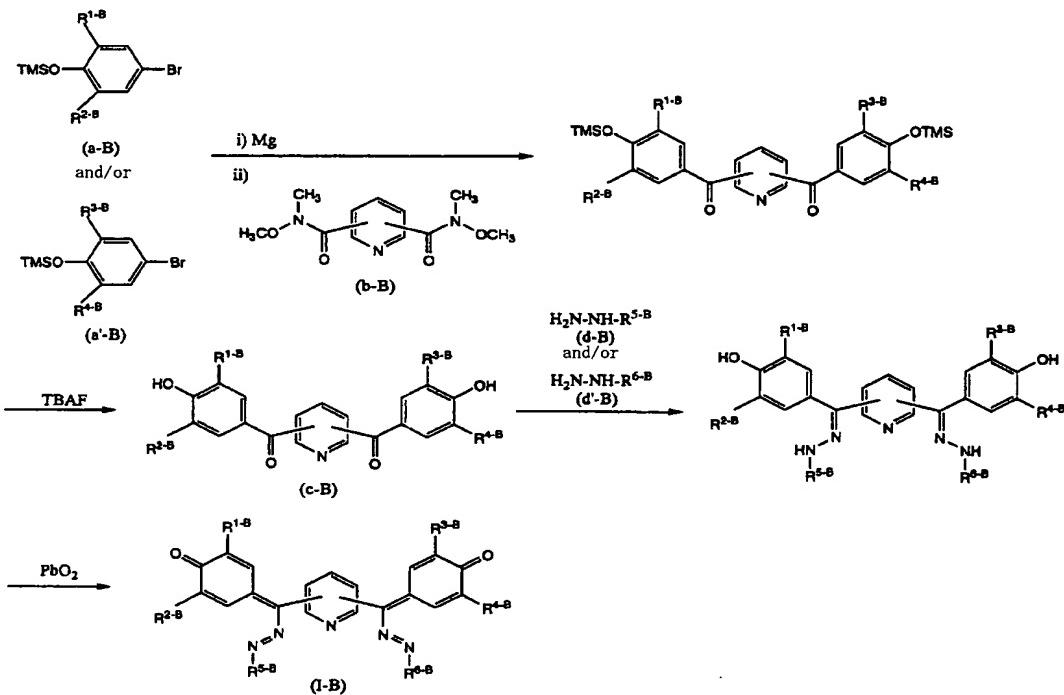
[0040]

Reaction scheme (1-B)



[0041]

Reaction scheme (2-B)



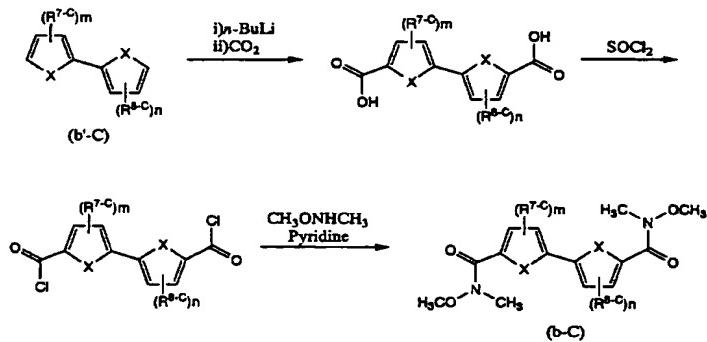
[0042]

In addition, the compound represented by the foregoing general formula (I-C) according to invention can be, for example, synthesized according to the following reaction schemes (1-C) and (2-C). That is, first of all, as shown in the following

reaction scheme (1-C), a compound represented by the structural formula (b'-C) is synthesized from a compound represented by the structural formula (b''-C). Next, as shown in the following reaction scheme (2-C), the compound represented by the structural formula (b-C) is allowed to react with a compound represented by the structural formula (a-C) and/or a compound represented by the structural formula (a''-C) and an appropriate organometallic reagent (for example, magnesium), and thereafter, a protective group (TMS: trimethylsilyl group) is eliminated off to synthesize a compound represented by the structural formula (c-C), followed by dehydration condensation with a compound represented by the structural formula (d-C) and/or a compound represented by the structural formula (d''-C) and subsequent oxidation with an appropriate catalyst (for example, lead dioxide ( $PbO_2$ )). Thus, the compound represented by the structural formula (I-C) can be synthesized. Incidentally, the term "TBAF" in the following reaction scheme (2-C) means tetrabutylammonium fluoride.

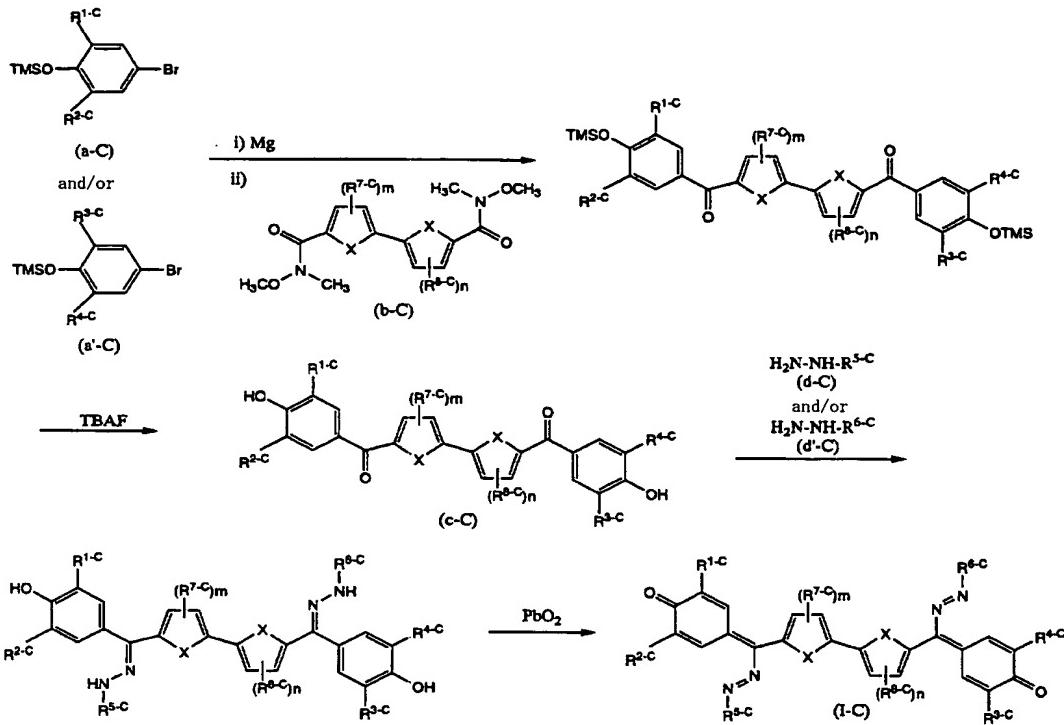
[0043]

Reaction scheme (1-C)



[0044]

Reaction scheme (2-C)



[0045]

Embodiments of the electrophotographic photoreceptor of the invention will be hereunder described with reference to the accompanying drawings.

Fig. 1 and Fig. 2 are each a schematic cross-sectional view to show a construction example of every kind of a photoreceptor. A reference numeral 1 stands for an electrically conductive substrate; reference numerals 2 and 5 stand for a photosensitive layer; a reference numeral 3 stands for a charge generation layer; a reference numeral 4 stands for a charge transport layer; and a reference numeral 6 stands for a covering layer, respectively. Fig. 1 shows one construction example of a so-called single layer type photoreceptor, in which the photosensitive layer 2 of a single layer containing a charge generation substance and a charge transport substance dispersed in a resin binder (binding agent) is provided on the electrically conductive substrate 1, and if desired, the covering layer (protective layer) 6 is laminated. This

single layer type photoreceptor can be prepared by dispersing a charge generation substance in a solution having a charge transport substance and a resin binder dissolved therein and coating this dispersion on an electrically conductive substrate. In addition, as the need arises, it is possible to coat and form the covering layer 6, for example, an inorganic metal protective layer.

[0046]

Fig. 2 shows one construction example of a so-called layered type photoreceptor, in which the photosensitive layer 5 in which the charge generation layer 3 composed mainly of a charge generation substance and the charge transport layer 4 containing a charge transport substance are successively laminated is provided on the electrically conductive substrate 1. This layered type photoreceptor can be prepared by vacuum depositing a charge generation substance on an electrically conductive substrate, or coating a dispersion obtained by dispersing a particle of a charge generation substance in a solvent or a resin binder on an electrically conductive substrate, followed by drying, to form a charge generation layer and subsequently coating thereon a solution having a charge transport substance and a resin binder dissolved therein, followed by drying to form a charge transport layer.

[0047]

Furthermore, in the photoreceptor of any type, it is possible to provide an undercoat layer between the electrically conductive substrate and the photosensitive layer, an aspect of which is, however, not illustrated. For the purpose of preventing injection of an unnecessary charge from the electrically conductive substrate into the photosensitive layer or enhancing covering of a defect on the substrate surface or adhesion of the photosensitive layer or the like, the undercoat layer can be provided depending upon the situation. The undercoat layer is comprised of a layer containing a resin as the major component or an oxidized film made of alumite or the like.

[0048]

Incidentally, any type of the photoreceptor of the invention contains, as the charge transport substance, at least one kind of the compound having electron transport properties according to the invention as represented by the foregoing general formula (1).

[0049]

A preferred embodiment of the photoreceptor of the invention will be hereunder described with reference to the layered type photoreceptor as shown in Fig. 2, but it should not be construed that the invention is limited to the following specific example.

[0050]

The electrically conductive substrate 1 serves as an electrode of the photoreceptor and simultaneously functions as a support for other respective layers, and may be in any of a cylindrical, plate-like or film-like form. With respect to the quality of the material, metals such as aluminum, stainless steel, and nickel, glasses or resins which have been subjected to an electrically conductive treatment, and the like can be employed.

[0051]

As described previously, the charge generation layer 3 is formed by coating a material of a dispersion having a particle of a charge generation substance in a resin binder or vacuum deposition, or other methods, and receives light to generate a charge. Also, it is important that its efficiency of charge generation is high, and at the same time, injection properties of the generated charge into the charge transport layer 4 are important. It is desired that the electric field dependency is low and that the injection is good even in a low electric field.

[0052]

As the charge generation substance, phthalocyanine compounds such as metal-free phthalocyanine and titanyl phthalocyanine, pigments or dyes such as a variety of azo, quinone, indigo, cyanine, squarylium, azulenium and pyrylium compounds, selenium or selenide compounds, and so on are employed. A suitable substance can be chosen depending upon the light wavelength region of an exposure light source to be used for the image formation. Since the charge generation layer is only required to have a charge generation function, its film thickness is determined by a coefficient of light absorption of the charge generation substance and is generally not more than 5  $\mu\text{m}$ , and suitably not more than 2  $\mu\text{m}$ . In addition, in the charge generation layer, it is also possible to add and use a charge transport substance, etc., with the charge generation substance being mainly contained.

[0053]

As the resin binder for the charge generation layer 3, polycarbonates, polyesters, polyamides, polyurethanes, vinyl chloride resins, phenoxy resins, polyvinyl butyral, diallyl phthalate resin, polymers or copolymers of methacrylic esters, and the like can be properly combined and used.

[0054]

The charge transport layer 4 is a coating film having a charge transport substance dispersed in a resin binder and exhibits a function to hold a charge of the photoreceptor as an insulator layer in a dark place and to transport a charge as injected from the charge generation layer at the time of receiving light. As described previously, in the invention, though it is required that at least one kind of the compound having electron transport properties according to the invention as represented by the foregoing general formula (I) is contained as such a charge transport substance, other charge transport substance may be contained. The amount of addition of the compound according to the invention is suitably from 10 to 60 % by

weight, and more suitably from 15 to 50 % by weight based on the whole of materials to be contained in the charge transport layer 4.

[0055]

As the resin binder for the charge transport layer 4, polycarbonates, polyesters, polystyrenes, polymers or copolymers of methacrylic esters, and the like can be used.

[0056]

For the purpose of preventing deterioration due to ozone, which becomes a hindrance in using the photoreceptor, from occurring or other purposes, it is also possible to contain, in the charge transport layer 4, an antioxidant such as amine based, phenol based, sulfur based, phosphite based, and phosphorus based antioxidants.

[0057]

The covering layer 6 as shown in Fig. 1 is required such that it has a function for receiving a charge of corona discharge and holding it in a dark place; that it has a performance for transmitting light to which the photosensitive layer is sensitive; and that at the time of exposure, it transmits light and makes it arrive at the photosensitive layer, and upon receipt of the injection of a generated charge, it neutralizes the surface charge, thereby making it extinct. As a material of the covering layer, an organic insulating film forming material such as polyesters and polyamides can be applied as an organic material. Furthermore, it is possible to use a mixture of such an organic material with an inorganic material such as glass and SiO<sub>2</sub> and further with a material capable of reducing electrical resistance such as metals and metal oxides. As the inorganic material, it is possible to apply a layer as prepared by film formation by the vapor phase epitaxy, such as amorphous silicon-carbon (SiC) composite film covering layers. It is desired that the covering layer is transparent as far as possible in a wavelength region of the absorption maximum of light of the charge generation substance as described previously.

[0058]

Though the film thickness of the covering layer itself varies depending upon the blending composition of the covering layer, it can be arbitrarily set within the range where at the time of repeated continuous use, adverse influences such as an increase of the residual potential are not revealed.

[0059]

Incidentally, even in the case of the single layer type photoreceptor as shown in Fig. 1, it is required to contain at least one kind of the compound having electron transport properties according to the invention as represented by the foregoing general formula (I) in the photosensitive layer 2. With respect to other materials, those which are the same as in the foregoing layered type photoreceptor can be used, and there are no particular limitations therein. Suitably, a hole transport substance is contained together with the compound of the foregoing general formula (I) as the charge transport substance. As the hole transport substance, benzidine derivatives, triphenylamine derivatives, and the like are preferable. In this case, with respect to the amounts of addition of these substances, the amount of addition of the compound according to the invention is suitably from 10 to 60 % by weight, and more preferably from 15 to 50 % by weight, and the amount of addition of the hole transport substance is suitably from 10 to 60 % by weight, and more suitably from 20 to 50 % by weight based on the whole of materials to be contained in the photosensitive layer forming coating film.

[0060]

Furthermore, in the surface layer of the photoreceptor (the covering layer in the case where the covering layer is provided, and the photosensitive layer in the outermost side in the case where the covering layer is not provided), it is possible to add a silicone oil as a leveling material; and for the purpose of imparting lubricating

properties, it is possible to contain a silicon or fluorine-containing polymer such as a silicone oil, a fluorine resin fine particle such as tetrafluoroethylene resins, a silicone resin fine particle, and a fluorine based comb type graft polymer.

## EXAMPLES

[0061]

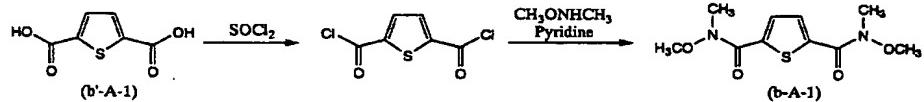
Examples of the invention will be hereunder described.

### Synthesis Example 1-A: Synthesis of compound of the foregoing specific example (I-A-3)

A compound represented by the foregoing specific example (I-A-3) was synthesized according to the following reaction schemes (1-A-1) and (2-A-1).

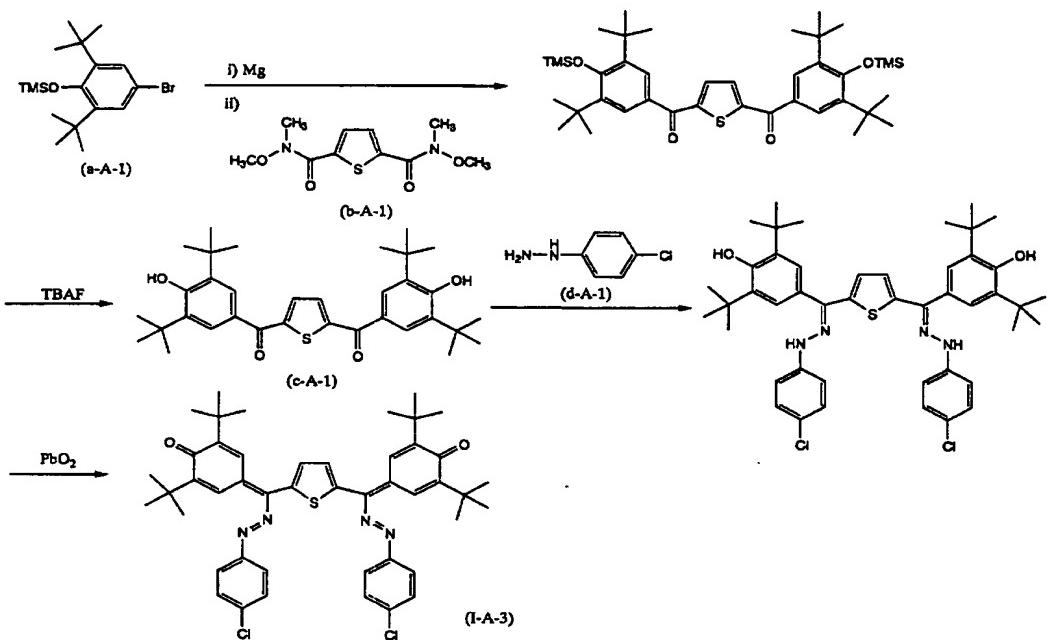
[0062]

#### Reaction scheme (1-A-1)



[0063]

Reaction scheme (2-A-1)



[0064]

To 100 mmoles (17.2 g) of thiophenedicarboxylic acid (the foregoing structural formula (b'-A-1)), 250 mmoles (29.8 g) of thionyl chloride was added, and the mixture was refluxed under heating. Thereafter, the excessive thionyl chloride was evaporated off in vacuo to obtain an oily substance. To a dichloromethane solution of this compound and 200 mmoles (19.5 g) of N,O-dimethylhydroxyamine hydrochloride, 600 mmoles (47.5 g) of pyridine was added under a nitrogen atmosphere at room temperature, and the mixture was stirred for 2 hours. Thereafter, the reaction solution was poured in aqueous hydrochloric acid and extracted with dichloromethane, followed by concentration. There was thus obtained, as a crude product, a compound represented by the foregoing structural formula (b-A-1) in a yield of 21.1 g (81.7 %).

[0065]

Next, to a THF solution of 30 mmoles (7.75 g) of the compound represented

by the foregoing structural formula (b-A-1), a Grignard reagent as prepared from 99 mmoles (2.41 g) of magnesium and a THF solution of 90 mmoles (32.2 g) of 4-bromo-2,6-di-t-butyl-1-(trimethylsiloxy)benzene (a-A-1) was added dropwise, and the mixture was stirred at room temperature for 3 hours. Thereafter, a small amount of a 1N hydrochloric acid aqueous solution was added to stop the reaction. In addition, 90 mmoles (90 mL) of a THF solution of 1.0 M tetrabutylammonium fluoride (TBAF) was added, followed by stirring. The reaction solution was poured in aqueous hydrochloric acid and extracted with dichloromethane, followed by concentration. There was thus obtained, as a crude product, a compound represented by the foregoing structural formula (c-A-1) in a yield of 8.12 g (49.2 %).

[0066]

In addition, 10 mmoles (5.48 g) of the compound represented by the foregoing structural formula (c-A-1) and 30 mmoles (5.37 g) of p-chlorophenylhydrazine hydrochloride (the foregoing structural formula (d-A-1)) were dissolved in a mixed solution of pyridine and ethanol, and the mixture was refluxed under heating. The reaction solution was poured in aqueous hydrochloric acid and extracted with dichloromethane, followed by concentration. Thereafter, the resulting product was purified by column chromatography to obtain a crude product.

[0067]

To a chloroform solution of the foregoing crude product, 30 mmoles (7.18 g) of lead dioxide ( $PbO_2$ ) was added at room temperature, and the mixture was stirred. After filtering off the residue, the reaction solution was concentrated, and the resulting solid was recrystallized from hexane to obtain a compound represented by the foregoing structural formula (I-A-3). Yield: 4.9 g (percent yield: 61.7 %), MS: m/z 792 ( $M^+$ ). Incidentally, the entire yield was 24.8 %. IR spectrum and  $^1H$ -NMR spectrum of this compound of the structural formula (I-A-3) are shown in Fig. 3 and

Fig. 4, respectively.

[0068]

Synthesis Example 2-A: Synthesis of compound of the foregoing specific example (I-A-5)

A compound represented by the foregoing structural formula (I-A-5) was obtained by following the same operations as in Synthesis Example 1-A, except for changing the 4-chlorophenylhydrazine as used in Synthesis Example 1-A to 3,5-dichlorophenylhydrazine. Entire yield: 18.9 %, MS: m/z 860 (M<sup>+</sup>). IR spectrum and <sup>1</sup>H-NMR spectrum of this compound of the structural formula (I-A-5) are shown in Fig. 5 and Fig. 6, respectively.

[0069]

Incidentally, the 4-bromo-2,6-di-tert-butyl-1-(trimethylsiloxy)benzene as used in the foregoing Synthesis Examples can be, for example, synthesized by a known method as described in the foregoing Patent Document 15 and so on.

[0070]

Photoreceptor Example 1-A

A plate-like photoreceptor for evaluating electrical characteristics and a drum-like photoreceptor for evaluating printing were prepared, respectively.

A solution for undercoat layer as prepared in the following manner was coated on the external surface of each of an aluminum plate (3 cm × 10 cm, thickness: 1 mm) and an aluminum drum (outer diameter: 30 mmΦ, length: 247.5 mm, thickness: 0.75 mm) by the dip coating method and then dried at 100°C for 60 minutes to remove the solvent. There was thus formed an undercoat layer having a film thickness of 0.3 μm.

[0071]

(Preparation of solution for undercoat layer)

- A-a1) Soluble nylon (AMILAN CM8000, manufactured by Toray Industries, Inc.): 3

parts by weight (30 g)

The foregoing material A-a1) for undercoat layer was stirred together with 97 parts by weight (970 g) of a mixed solvent of methanol/methylene chloride (5 vol./5 vol.) and dissolved to prepare a solution for undercoat layer.

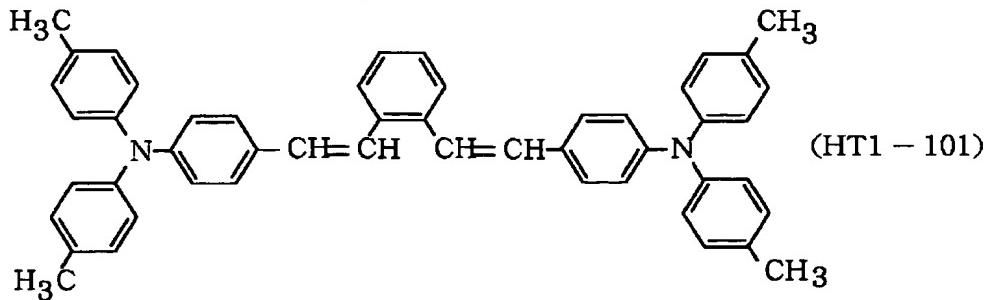
[0072]

Next, on this undercoat layer, a dispersion for single layer type photosensitive layer as prepared in the following manner was coated by the dip coating method for the plate-like photoreceptor and by the ring coating method for the drum-like photoreceptor, respectively and then dried at 100°C for 60 minutes to remove the solvent. Thus, a single layer type photosensitive layer having a film thickness of 25 µm was formed, from which was then prepared an electrophotographic photoreceptor.

[0073]

(Preparation of dispersion for single layer type photosensitive layer)

- A-b1) Charge generation substance: X type metal-free phthalocyanine (see Fig. 2 of JP-A-2001-228637): 0.3 parts by weight (0.15 g)
- A-b2) Hole transport substance: Triphenylamine compound represented by the following structural formula (HT1-101) ((HT1-101) of JP-A-2000-314969): 7 parts by weight (3.5 g)



- A-b3) Electron transport substance: Compound represented by the foregoing structural formula (I-A-3): 3 parts by weight (1.5 g)
- A-b4) Antioxidant: 3,5-Di-tert-4-hydroxytoluene (BHT) (manufactured by Tokyo

Kasei Kogyo Co., Ltd.): 1 part by weight (0.5 g)

- A-b5) Silicone oil: KF-50 (manufactured by Shin-Etsu Chemical Co., Ltd.): 0.01 parts by weight (0.005 g)
- A-b6) Binder resin: Bisphenol Z type polycarbonate resin (PANLITE TS2020, manufactured by Teijin Chemicals Ltd.) ((BD1-1) of JP-A-2000-314969): 10 parts by weight (5 g)

[0074]

In a 100-mL plastic bottle, the foregoing materials A-b1) to A-b6) for photo-sensitive layer were charged together with 100 parts by weight (50 g) of a methylene chloride solvent and 50 g of stainless steel beads (3 mmφ), and the mixture was subjected to a dispersing treatment for 60 minutes by a paint conditioner Model 5400 (Red Devil Equipment Co., U.S.A.). Thereafter, the stainless steel beads were separated to prepare a dispersion for single layer type photosensitive layer.

[0075]

#### Photoreceptor Example 2-A

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-A, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-A, 3 parts by weight of the compound represented by the foregoing structural formula (I-A-3) as the electron transport substance was replaced by 3 parts by weight of the compound represented by the foregoing structural formula (I-A-5).

[0076]

#### Photoreceptor Example 3-A

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-A, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-A, 3 parts by weight of the

compound represented by the foregoing structural formula (I-A-3) as the electron transport substance was replaced by 3 parts by weight of the compound represented by the foregoing structural formula (I-A-26).

[0077]

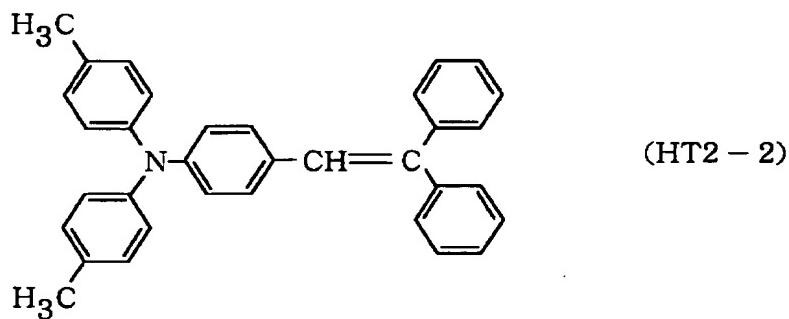
Photoreceptor Example 4-A

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-A, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-A, 3 parts by weight of the compound represented by the foregoing structural formula (I-A-3) as the electron transport substance was replaced by 3 parts by weight of the compound represented by the foregoing structural formula (I-A-30).

[0078]

Photoreceptor Example 5-A

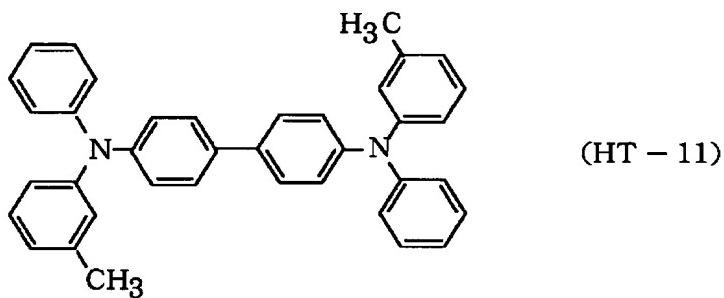
A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-A, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-A, 7 parts by weight of the triphenylamine compound represented by the foregoing structural formula (HT1-101) as the hole transport substance was replaced by 7 parts by weight of a styryl compound represented by the following structural formula (HT2-2) ((HT2-2) of JP-A-2000-314969).



[0079]

Photoreceptor Example 6-A

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-A, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-A, 7 parts by weight of the triphenylamine compound represented by the foregoing structural formula (HT1-101) as the hole transport substance was replaced by 7 parts by weight of a benzidine derivative represented by the following structural formula (HT-11) ((HT-11) of JP-A-2000-314969).



[0080]

Photoreceptor Example 7-A

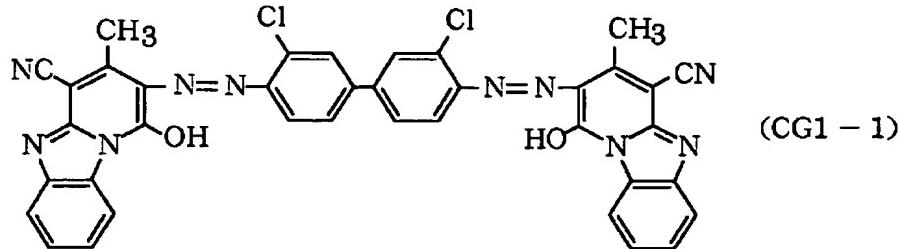
A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-A, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-A, 0.3 parts by weight of the X type metal-free phthalocyanine as the charge generation substance was replaced by 0.2 parts by weight of Y type titanyl phthalocyanine (see Fig. 4 of JP-A-2001-228637).

[0081]

Photoreceptor Example 8-A

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-A, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-A, 0.3 parts by weight of the

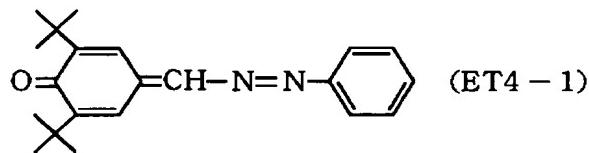
X type metal-free phthalocyanine as the charge generation substance was replaced by a bisazo pigment represented by the following formula (CG1-1).



[0082]

Photoreceptor Comparative Example 1-A

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-A, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-A, 3 parts by weight of the compound represented by the foregoing structural formula (I-A-3) as the electron transport substance was replaced by 3 parts by weight of a compound represented by the following structural formula (ET4-1) ((ET4-1) of JP-A-2000-314969).



[0083]

Evaluation of photoreceptors of Photoreceptor Examples 1-A to 8-A and Photoreceptor Comparative Example 1

Using a plate-like photoreceptor for evaluating electrical characteristics, the evaluation was carried out by an electrostatic copying paper analyzer EPA-8100 as manufactured by Kawaguchi Electric Works Co., Ltd.

First of all, the photoreceptor was charged in a dark place under an atmosphere at a temperature of 23°C and a humidity of 45 % such that the surface

potential became about +700 V, and a retention rate  $V_{k5}$  of the surface potential after a lapse of 5 seconds was determined according to the following expression.

$$\text{Retention rate } V_{k5} (\%) = (V_5/V_0) \times 100$$

$V_0$ : Surface potential immediately after charging

$V_5$ : Surface potential 5 seconds after charging

[0084]

Next, the surface potential was changed to +600 V, monochromic light of 1.0  $\mu\text{W}/\text{cm}^2$  resulting from separation of light of a halogen lamp into 780 nm (but into 550 nm with respect to the Photoreceptor Example 8-A) was exposed for 5 seconds, the amount of exposure required until the surface potential became one-half (+300 V) was determined as a sensitivity  $E_{1/2}$  ( $\mu\text{J}/\text{cm}^2$ ), and the surface potential 5 seconds after the exposure was determined as a residual potential  $V_r$  (V).

The evaluation results are shown in the following Table 1.

[0085]

Table 1

	Retention rate $V_{k5}$ (%)	Sensitivity $E_{1/2}$ ( $\mu\text{J}/\text{cm}^2$ )	Residual potential $V_r$ (V)
Photoreceptor Example 1-A	85.6	0.29	24
Photoreceptor Example 2-A	82.2	0.37	31
Photoreceptor Example 3-A	83.5	0.35	33
Photoreceptor Example 4-A	84.4	0.40	39
Photoreceptor Example 5-A	81.5	0.32	36
Photoreceptor Example 6-A	80.9	0.33	38
Photoreceptor Example 7-A	83.9	0.31	28
Photoreceptor Example 8-A *)	82.7	0.49	54
Photoreceptor Comparative Example 1-A	80.3	0.53	84

\*) Light at 550 nm was employed for exposure only in Example 8-A.

[0086]

Furthermore, a drum-like photoreceptor (30 mmφ) was used for evaluating the durability by actual printing. Each drum-like photoreceptor was installed in a laser printer HL-1240 as manufactured by Brother Industries, Ltd. and printed with a black solid image, a white solid image, and a half-tone image, respectively under an atmosphere at a temperature of 22°C and a humidity of 44 %. Subsequently, 5,000 sheets were printed with an image having a rate of printing of about 5 %. Thereafter, a black solid image, a white solid image, and a half-tone image were again printed, respectively, thereby evaluating the image after printing 5,000 sheets.

[0087]

As a result, in the photoreceptors of Photoreceptor Examples 1-A to 7-A, a good image was obtained in both the initial image and the image after printing 5,000 sheets.

[0088]

Incidentally, with respect to Photoreceptor Example 8-A, the photoreceptor did not have sufficient sensitivity in a laser wavelength region (in the vicinity of 780 nm) of the used laser printer and therefore, it was noted that this photoreceptor is not suitable for this laser printer.

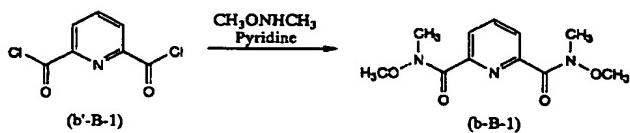
[0089]

Synthesis Example 1-B: Synthesis of compound of the foregoing specific example (I-B-4)

A compound of the foregoing specific example (I-B-4) was synthesized according to the following reaction schemes (1-B-1) and (2-B-1).

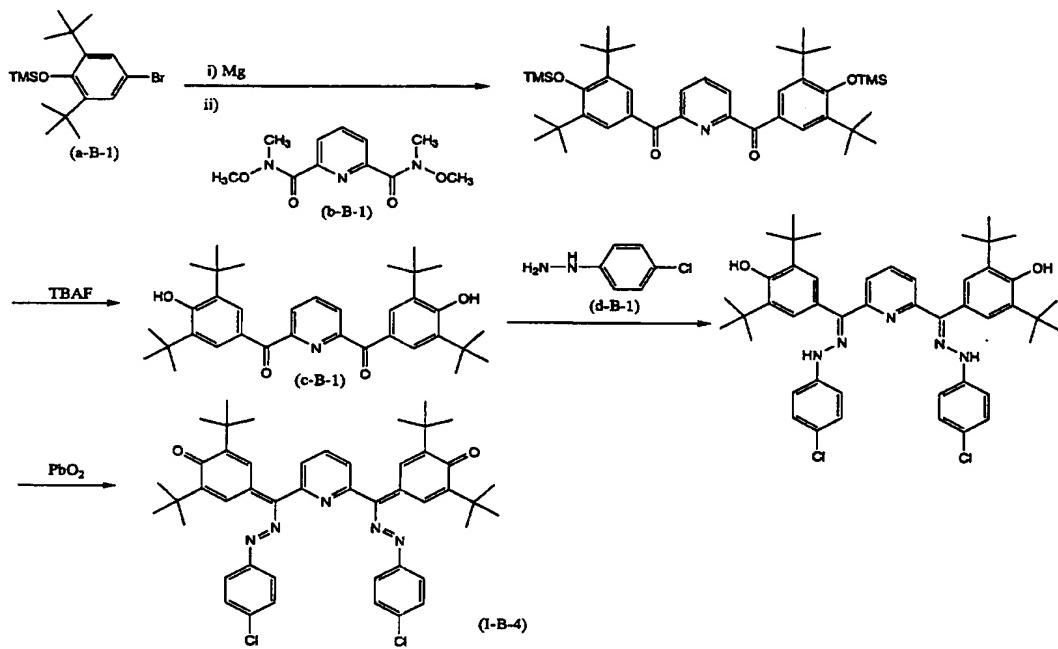
[0090]

Reaction scheme (1-B-1)



[0091]

Reaction scheme (2-B-1)



[0092]

In a three-necked flask, 480 mmoles (38.0 g) of pyridine was added in a dichloromethane solution of 100 mmoles (20.4 g) of 2,6-pyridinecarbonyl dichloride (the foregoing structural formula (b'-B-1)) and 230 mmoles (22.4 g) of N,O-dimethylhydroxyamine hydrochloride at room temperature under a nitrogen atmosphere, and the mixture was stirred for 2 hours. Thereafter, the reaction solution was poured in aqueous hydrochloric acid and extracted with dichloromethane, followed by concentration. There was thus obtained, as a crude product, a compound represented by the foregoing structural formula (b-B-1) in a yield of 22.8 g (89.9 %).

[0093]

Next, to a THF solution of 30 mmoles (7.6 g) of the compound represented by the foregoing structural formula (b-B-1), a Grignard reagent as prepared from 99 mmoles (2.4 g) of magnesium and a THF solution of 90 mmoles (32.2 g) of 4-bromo-2,6-di-t-butyl-1-(trimethylsiloxy)benzene (the foregoing structural formula (a-B-1)) was added dropwise, and the mixture was stirred at room temperature for 3 hours. Thereafter, a small amount of a 1N hydrochloric acid aqueous solution was added to stop the reaction. In addition, 90 mmoles (90 mL) of a THF solution of 1.0 M tetrabutylammonium fluoride (TBAF) was added, followed by stirring. The reaction solution was poured in aqueous hydrochloric acid and extracted with dichloromethane, followed by concentration. There was thus obtained, as a crude product, a compound represented by the foregoing structural formula (c-B-1) in a yield of 12.8 g (78.5 %).

[0094]

In addition, 10 mmoles (5.4 g) of the compound represented by the foregoing structural formula (c-B-1) and 30 mmoles (5.4 g) of 4-chlorophenylhydrazine (the foregoing structural formula (d-B-1)) were dissolved in pyridine, and the mixture was refluxed under heating. The reaction solution was poured in aqueous hydrochloric acid and extracted with dichloromethane, followed by concentration. Thereafter, the resulting product was purified by column chromatography to obtain a crude product.

[0095]

To a chloroform solution of the foregoing crude product, 30 mmoles (7.2 g) of lead dioxide ( $PbO_2$ ) was added at room temperature, and the mixture was stirred. After filtering off the residue, the reaction solution was concentrated, and the resulting solid was recrystallized from hexane to obtain a compound represented by the foregoing structural formula (I-B-4). Yield: 3.3 g (percent yield: 41.8 %), MS: m/z 787 ( $M^+$ ). Incidentally, the entire yield was 29.5 %. IR spectrum and  $^1H$ -NMR

spectrum of this compound of the structural formula (I-B-4) are shown in Fig. 7 and Fig. 8, respectively.

[0096]

Incidentally, the 4-bromo-2,6-di-tert-butyl-1-(trimethylsiloxy)benzene (the foregoing structural formula (a-B-1)) as used above can be, for example, synthesized by a known method as described in JP-A-2001-222122 and so on.

[0097]

Synthesis Example 2-B: Synthesis of compound of the foregoing specific example (I-B-7)

A compound represented by the foregoing structural formula (I-B-7) was obtained by following the same operations as in the foregoing Synthesis Example 1-B, except for changing the 4-chlorophenylhydrazine to 2,4-dichlorophenylhydrazine. Entire yield: 25.9 %, MS: m/z 855 (M<sup>+</sup>). IR spectrum and <sup>1</sup>H-NMR spectrum of this compound of the specific example (I-B-7) are shown in Fig. 9 and Fig. 10, respectively.

[0098]

Synthesis Example 3-B: Synthesis of compound of the foregoing specific example (I-B-12)

A compound represented by the foregoing structural formula (I-B-12) was obtained by following the same operations as in the foregoing Synthesis Example 1-B, except for changing the 4-chlorophenylhydrazine to 2,4,6-trichlorophenylhydrazine. Entire yield: 46.1 %, MS: m/z 923 (M<sup>+</sup>). IR spectrum and <sup>1</sup>H-NMR spectrum of this compound of the specific example (I-B-12) are shown in Fig. 11 and Fig. 12, respectively.

[0099]

Photoreceptor Example 1-B

A plate-like photoreceptor for evaluating electrical characteristics and a drum-like photoreceptor for evaluating printing were prepared, respectively.

A solution for undercoat layer as prepared in the following manner was coated on the external surface of each of an aluminum plate (3 cm × 10 cm, thickness: 1 mm) and an aluminum drum (outer diameter: 30 mmφ, length: 247.5 mm, thickness: 0.75 mm) by the dip coating method and then dried at 100°C for 60 minutes to remove the solvent. There was thus formed an undercoat layer having a film thickness of 0.3 μm.

[0100]

(Preparation of solution for undercoat layer)

- B-a1) Soluble nylon (AMILAN CM8000, manufactured by Toray Industries, Inc.): 3 parts by weight (30 g)

The foregoing material B-a1) for undercoat layer was stirred together with 97 parts by weight (970 g) of a mixed solvent of methanol/methylene chloride (5 vol./5 vol.) and dissolved to prepare a solution for undercoat layer.

[0101]

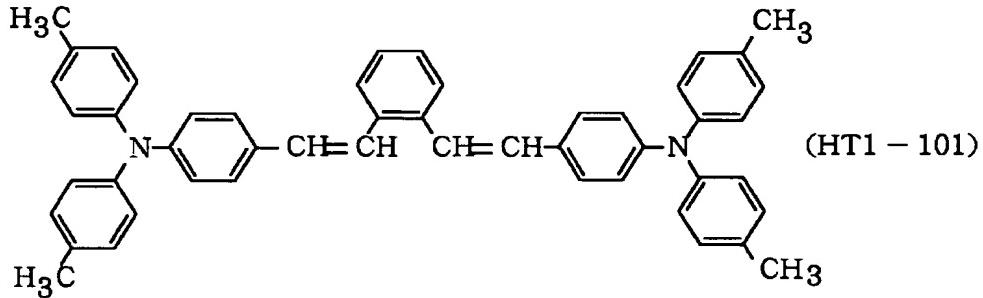
Next, on this undercoat layer, a dispersion for single layer type photosensitive layer as prepared in the following manner was coated by the dip coating method for the plate-like photoreceptor and by the ring coating method for the drum-like photoreceptor, respectively and then dried at 100°C for 60 minutes to remove the solvent. Thus, a single layer type photosensitive layer having a film thickness of 25 μm was formed, from which was then prepared an electrophotographic photoreceptor.

[0102]

(Preparation of dispersion for single layer type photosensitive layer)

- B-b1) Charge generation substance: X type metal-free phthalocyanine (see Fig. 2 of JP-A-2001-228637): 0.3 parts by weight (0.15 g)
- B-b2) Hole transport substance: Triphenylamine compound represented by the

following structural formula (HT1-101) ((HT1-101) of JP-A-2000-314969): 7 parts by weight (3.5 g)



- B-b3) Electron transport substance: Compound represented by the foregoing structural formula (I-B-4): 2 parts by weight (1 g)
- B-b4) Antioxidant: 3,5-Di-tert-4-hydroxytoluene (BHT) (manufactured by Tokyo Kasei Kogyo Co., Ltd.): 1 part by weight (0.5 g)
- B-b5) Silicone oil: KF-50 (manufactured by Shin-Etsu Chemical Co., Ltd.): 0.01 parts by weight (0.005 g)
- B-b6) Binder resin: Bisphenol Z type polycarbonate resin (PANLITE TS2020, manufactured by Teijin Chemicals Ltd.) ((BD1-1) of JP-A-2000-314969): 11 parts by weight (5.5 g)

[0103]

In a 100-mL plastic bottle, the foregoing materials B-b1) to B-b6) for photo-sensitive layer were charged together with 100 parts by weight (50 g) of a methylene chloride solvent and 50 g of stainless steel beads (3 mmφ), and the mixture was subjected to a dispersing treatment for 60 minutes by a paint conditioner Model 5400 (Red Devil Equipment Co., U.S.A.). Thereafter, the stainless steel beads were separated to prepare a dispersion for single layer type photosensitive layer.

[0104]

Photoreceptor Example 2-B

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-B, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-B, 2 parts by weight of the compound represented by the foregoing structural formula (I-B-4) as the electron transport substance was replaced by 2 parts by weight of the compound represented by the foregoing structural formula (I-B-7).

[0105]

Photoreceptor Example 3-B

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-B, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-B, 2 parts by weight of the compound represented by the foregoing structural formula (I-B-4) as the electron transport substance was replaced by 2 parts by weight of the compound represented by the foregoing structural formula (I-B-12).

[0106]

Photoreceptor Example 4-B

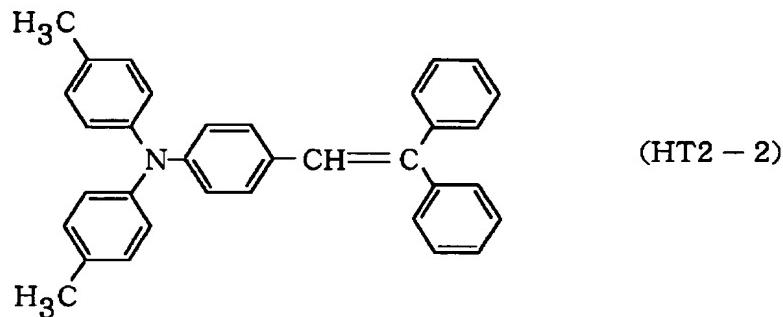
A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-B, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-B, 0.3 parts by weight of the X type metal-free phthalocyanine as the charge generation substance was replaced by 0.2 parts by weight of Y type titanyl phthalocyanine (see Fig. 4 of JP-A-2001-228637).

[0107]

Photoreceptor Example 5-B

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-B, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-B, 7 parts by weight of the

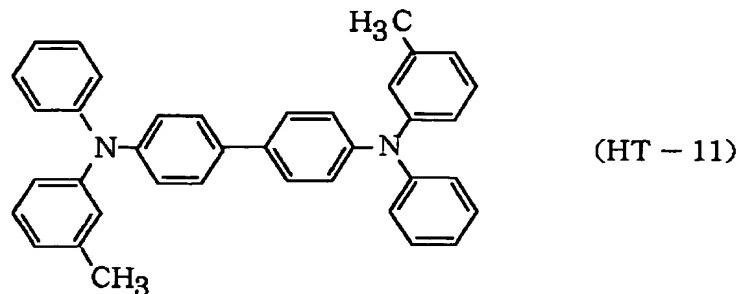
triphenylamine compound represented by the foregoing structural formula (HT1-101) as the hole transport substance was replaced by 7 parts by weight of a styryl compound represented by the following structural formula (HT2-2) ((HT2-2) of JP-A-2000-314969).



[0108]

Photoreceptor Example 6-B

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-B, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-B, 7 parts by weight of the triphenylamine compound represented by the foregoing structural formula (HT1-101) as the hole transport substance was replaced by 7 parts by weight of a benzidine derivative represented by the following structural formula (HT-11) ((HT-11) of JP-A-2000-314969).



[0109]

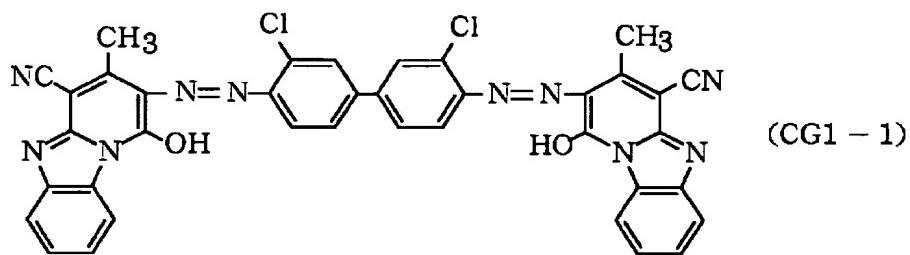
### Photoreceptor Example 7-B

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-B, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-B, 0.3 parts by weight of the X type metal-free phthalocyanine as the charge generation substance was replaced by 0.2 parts by weight of amorphous titanyl phthalocyanine (see Fig. 5 of JP-A-2001-228637).

[0110]

### Photoreceptor Example 8-B

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-B, except that in the composition of the dispersion for photosensitive layer as used in Photoreceptor Example 1-B, 0.3 parts by weight of the X type metal-free phthalocyanine as the charge generation substance was replaced by a bisazo pigment represented by the following formula (CG1-1).



[0111]

### Evaluation of Photoreceptor Examples 1-B to 8-B

Using a plate-like photoreceptor for evaluating electrical characteristics, the evaluation was carried out by an electrostatic copying paper analyzer EPA-8100 as manufactured by Kawaguchi Electric Works Co., Ltd.

First of all, the photoreceptor was charged in a dark place under an atmosphere at a temperature of 23°C and a humidity of 45 % such that the surface

potential became about +700 V, and a retention rate  $V_{k5}$  of the surface potential after a lapse of 5 seconds was determined according to the following expression.

$$\text{Retention rate } V_{k5} (\%) = (V_5/V_0) \times 100$$

$V_0$ : Surface potential immediately after charging

$V_5$ : Surface potential 5 seconds after charging

[0112]

Next, the surface potential was changed to +600 V, monochromic light of 1.0  $\mu\text{W}/\text{cm}^2$  resulting from separation of light of a halogen lamp into 780 nm (but into 550 nm with respect to the Photoreceptor Example 8-B) was exposed for 5 seconds, the amount of exposure required until the surface potential became one-half (+300 V) was determined as a sensitivity  $E_{1/2}$  ( $\mu\text{J}/\text{cm}^2$ ), and the surface potential 5 seconds after the exposure was determined as a residual potential  $V_r$  (V).

The evaluation results are shown in the following Table 2.

[0113]

Table 2

		Retention rate $V_{k5}$ (%)	Sensitivity $E_{1/2}$ ( $\mu J/cm^2$ )	Residual potential $V_r$ (V)
Photoreceptor 1-B	Example	89.3	0.53	60
Photoreceptor 2-B	Example	87.2	0.56	58
Photoreceptor 3-B	Example	86.1	0.53	70
Photoreceptor 4-B	Example	85.8	0.46	54
Photoreceptor 5-B	Example	88.7	0.49	58
Photoreceptor 6-B	Example	82.4	0.50	61
Photoreceptor 7-B	Example	84.3	0.52	54
Photoreceptor 8-B *)	Example	79.7	0.61	69

\*) Light at 550 nm was employed for exposure only in Example 8-B.

[0114]

Furthermore, a drum-like photoreceptor (30 mmΦ) was used for evaluating the durability by actual printing. Each drum-like photoreceptor was installed in a laser printer HL-1240 as manufactured by Brother Industries, Ltd. and printed with a black solid image, a white solid image, and a half-tone image, respectively under an atmosphere at a temperature of 24°C and a humidity of 53 %. Subsequently, 5,000 sheets were printed with an image having a rate of printing of about 5 %. Thereafter, a black solid image, a white solid image, and a half-tone image were again printed, respectively, thereby evaluating the image after printing 5,000 sheets.

[0115]

As a result, in the photoreceptors of Photoreceptor Examples 1-B to 7-B, a good image was obtained in both the initial image and the image after printing 5,000 sheets.

[0116]

Incidentally, with respect to Photoreceptor Example 8-B, the photoreceptor did not have sufficient sensitivity in a laser wavelength region (in the vicinity of 780 nm) of the used laser printer and therefore, it was noted that this photoreceptor is not suitable for this laser printer.

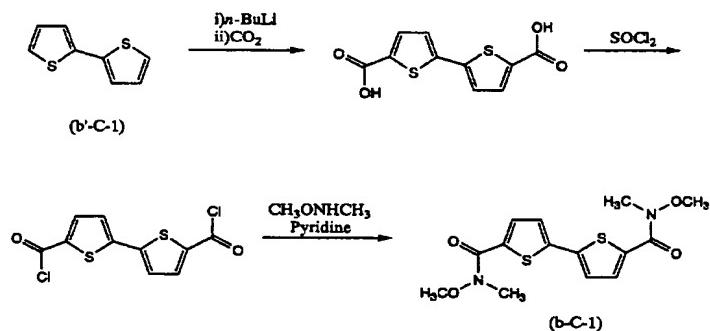
[0117]

Synthesis Example 1-C: Synthesis of compound of the foregoing specific example (I-C-1)

A compound of the foregoing specific example (I-C-1) was synthesized according to the following reaction schemes (1-C-1) and (2-C-1).

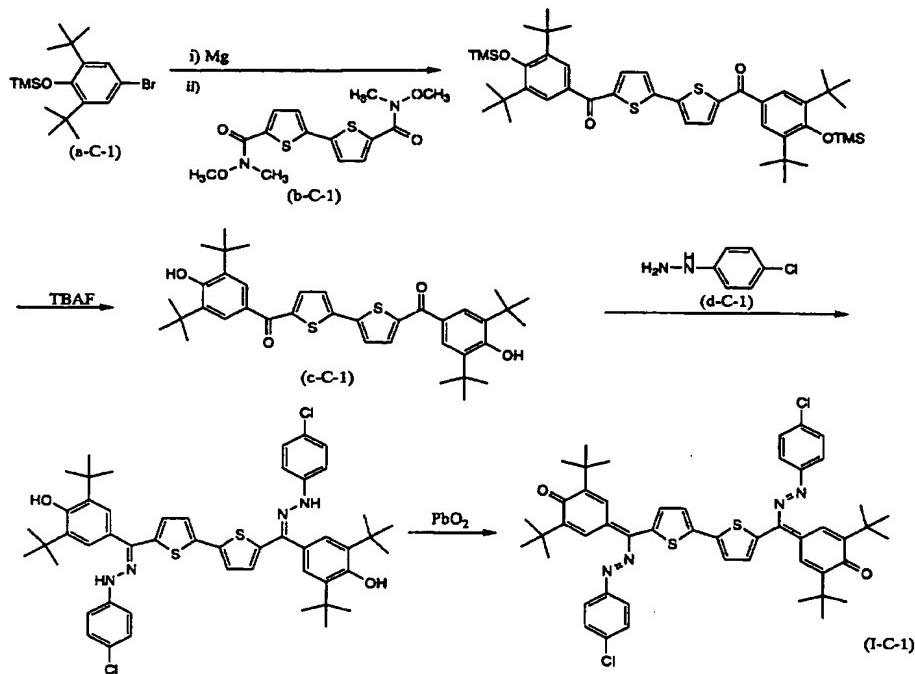
[0118]

Reaction scheme (1-C-1)



[0119]

Reaction scheme (2-C-1)



[0120]

To a hexane solution of 100 mmoles (16.6 g) of bithiophene and 220 mmoles (25.6 g) of N,N,N,N-tetramethylethylenediamine, 220 mmoles (138 mL) of n-BuLi was added dropwise in an ice bath under a nitrogen atmosphere. Thereafter, the mixture was agitated at room temperature for 0.5 hours and additionally refluxed under heating for 0.5 hours. The reaction solution was cooled and then poured in dry ice, followed by allowing it to stand overnight. The resulting reaction solution was extracted with chloroform and a 10 % sodium hydroxide aqueous solution. When the resulting aqueous solution was made acidic with hydrochloric acid, a precipitate was deposited. By performing filtration, bithiophenedicarboxylic acid was obtained as a crude product in a yield of 19.3 g (percent yield: 75.9 %).

[0121]

To 50 mmoles (12.7 g) of the resulting bithiophenedicarboxylic acid, 150 mmoles (17.8 g) of thionyl chloride was added, and three drops of N,N-dimethylformamide was added dropwise, followed by refluxing under heating.

After a lapse of 2 hours, the excessive thionyl chloride was evaporated off to obtain a crude product. To a dichloromethane solution of this compound and 110 mmoles (10.7 g) of N,O-dimethylhydroxyamine hydrochloride, 330 mmoles (26.1 g) of pyridine was added under a nitrogen atmosphere at room temperature, and the mixture was stirred for 2 hours. Thereafter, the reaction solution was poured in aqueous hydrochloric acid and extracted with dichloromethane, followed by concentration. There was thus obtained, as a crude product, a compound represented by the foregoing structural formula (b-C-1) in a yield of 8.2 g (48.2 %).

[0122]

Next, to a THF solution of 10 mmoles (3.4 g) of the compound represented by the foregoing structural formula (b-C-1), a Grignard reagent as prepared from 36 mmoles (0.9 g) of magnesium and a THF solution of 30 mmoles (10.7 g) of 4-bromo-2,6-di-t-butyl-1-(trimethylsiloxy)benzene (the foregoing structural formula (a-C-1)) was added dropwise, and the mixture was stirred at room temperature for 3 hours. Thereafter, a small amount of a 1N hydrochloric acid aqueous solution was added to stop the reaction. In addition, 30 mmoles (30 mL) of a THF solution of 1.0 M tetrabutylammonium fluoride (TBAF) was added, followed by stirring. The reaction solution was poured in aqueous hydrochloric acid and extracted with dichloromethane, followed by concentration. There was thus obtained, as a crude product, a compound represented by the foregoing structural formula (c-C-1) in a yield of 2.6 g (41.7 %).

[0123]

In addition, 4 mmoles (2.5 g) of the compound represented by the foregoing structural formula (c-C-1) and 40 mmoles (7.2 g) of 4-chlorophenylhydrazine hydrochloride (the foregoing structural formula (d-C-1)) were dissolved in pyridine, and the mixture was refluxed under heating. The reaction solution was poured in

aqueous hydrochloric acid and extracted with dichloromethane, followed by concentration. Thereafter, the resulting product was purified by column chromatography to obtain a crude product.

[0124]

To a chloroform solution of the foregoing crude product, 6.9 mmoles (1.7 g) of lead dioxide ( $PbO_2$ ) was added at room temperature, and the mixture was stirred. After filtering off the residue, the reaction solution was concentrated, and the resulting solid was recrystallized from a mixed solution of chloroform and hexane to obtain a compound represented by the foregoing structural formula (I-C-1). Yield: 2.0 g (percent yield: 57.1 %), MS:  $m/z$  874 ( $M^+$ ). Incidentally, the entire yield was 8.7 %. IR spectrum and  $^1H$ -NMR spectrum of this compound of the structural formula (I-C-1) are shown in Fig. 13 and Fig. 14, respectively.

[0125]

Incidentally, the 4-bromo-2,6-di-tert-butyl-1-(trimethylsiloxy)benzene (the foregoing structural formula (a-C-1)) as used in the foregoing Synthesis Example can be, for example, synthesized by a known method as described in JP-A-2001-222122 and so on.

[0126]

#### Photoreceptor Example 1-C

20 parts by weight (2 g) of X type metal-free phthalocyanine ( $H_2Pc$ ) and 100 parts by weight (10 g) of the compound represented by the foregoing structural formula (I-C-1) were kneaded together with 100 parts by weight (10 g) of a polyester resin (trade name: VYLON 200, manufactured by Toyobo Co., Ltd.) and a tetrahydrofuran (THF) solvent for 3 hours by using a mixer to prepare a coating liquid. The coating liquid was coated on an aluminum-made drum having an outer diameter of 30 mm and a length of 260 mm as an electrically conductive substrate to form a

photosensitive layer in a film thickness after drying of 15  $\mu\text{m}$ , from which was then prepared a photoreceptor.

[0127]

Photoreceptor Example 2-C

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-C, except that in Photoreceptor Example 1-C, the compound represented by the foregoing structural formula (I-C-3) was used in place of the compound represented by the foregoing structural formula (I-C-1).

[0128]

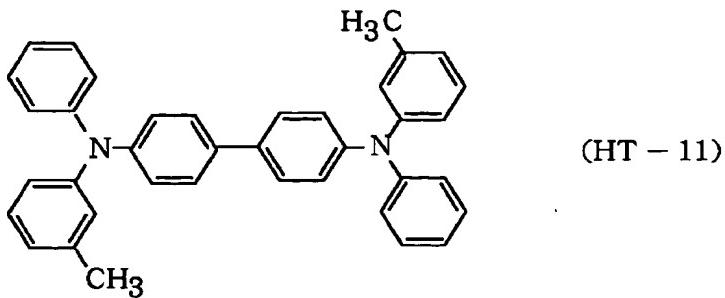
Photoreceptor Example 3-C

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 1-C, except that in Photoreceptor Example 1-C, the compound represented by the foregoing structural formula (I-C-5) was used in place of the compound represented by the foregoing structural formula (I-C-1).

[0129]

Photoreceptor Example 4-C

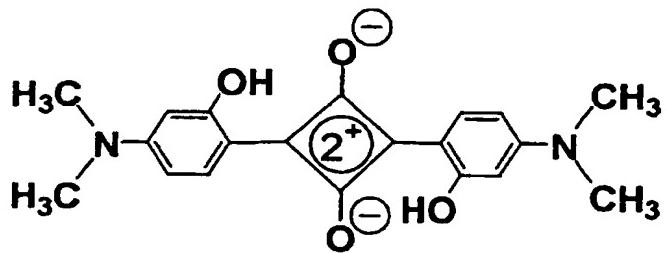
2 parts by weight (0.2 g) of X type metal-free phthalocyanine ( $\text{H}_2\text{Pc}$ ), 40 parts by weight (4 g) of the compound represented by the structural formula (I-C-7), 60 parts by weight (6 g) of a benzidine derivative represented by the following structural formula (HT-11) as a hole transport substance, and 100 parts by weight (10 g) of a polycarbonate resin (trade name: PCZ-200, manufactured by Mitsubishi Gas Chemical Company, Inc.) were kneaded together with methylene chloride for 3 hours by using a mixer to prepare a coating liquid. The coating liquid was coated on the same aluminum-made drum as in Example 1-C to form a photosensitive layer in a film thickness after drying of 25  $\mu\text{m}$ , from which was then prepared a single layer type photoreceptor.



[0130]

Photoreceptor Example 5-C

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 4-C, except that in Photoreceptor Example 4-C, a squarylium compound represented by the following structural formula was used in place of the X type metal-free phthalocyanine and that the compound represented by the foregoing structural formula (I-C-11) was used in place of the compound represented by the foregoing structural formula (I-C-7).



[0131]

Photoreceptor Example 6-C

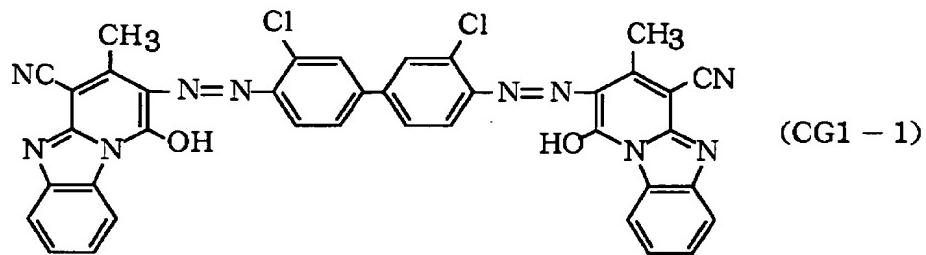
70 parts by weight (7 g) of  $\alpha$ -type titanyl phthalocyanine (TiOPc) and 30 parts by weight (3 g) of a vinyl chloride copolymer (trade name: MR-110, manufactured by Zeon Corporation) were kneaded together with methylene chloride for 3 hours by using a mixer to prepare a coating liquid. The coating liquid was coated on the same aluminum-made drum as in Example 1-C to form a charge generation layer in a film

thickness after drying of 1  $\mu\text{m}$ . Next, 100 parts by weight (10 g) of the compound represented by the foregoing structural formula (I-C-1), 100 parts by weight (10 g) of a polycarbonate resin (trade name: PCZ-200, manufactured by Mitsubishi Gas Chemical Company, Inc.), and 0.1 parts by weight (0.01 g) of silicone oil were mixed with methylene chloride. The mixture was coated on the charge generation layer to form a charge transport layer in a film thickness after drying of 7  $\mu\text{m}$ . There was thus prepared a layered type photoreceptor.

[0132]

Photoreceptor Example 7-C

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 4-C, except that in Photoreceptor Example 4-C, a bisazo pigment represented by the following structural formula (CG1-1) was used in place of the X type metal-free phthalocyanine and that the compound represented by the foregoing structural formula (I-C-6) was used in place of the compound represented by the foregoing structural formula (I-C-7).

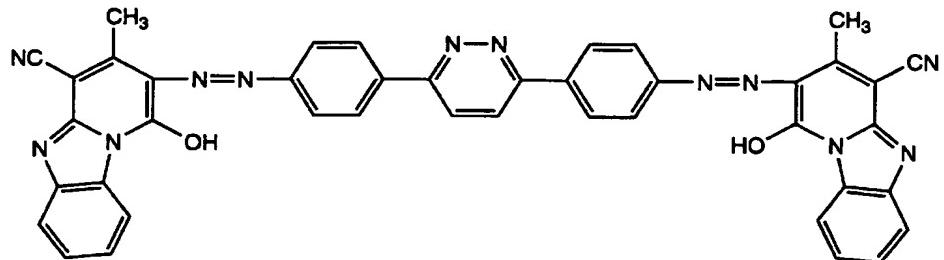


[0133]

Photoreceptor Example 8-C

A single layer type photoreceptor was prepared in the same manner as in Photoreceptor Example 4-C, except that in Photoreceptor Example 4-C, a bisazo pigment represented by the following structural formula was used in place of the X type metal-free phthalocyanine and that the compound represented by the foregoing

structural formula (I-C-12) was used in place of the compound represented by the foregoing structural formula (I-C-7).



[0134]

Electrophotographic characteristics of each of the thus obtained photoreceptors of the Examples were measured.

An initial surface potential when the surface of the photoreceptor was positively charged by performing corona discharge of +4.5 kV in a dark place was designated as  $V_s$  (V), and a surface potential  $V_d$  (V) when the photoreceptor was subsequently held in a dark place for 5 seconds in the state that the corona discharge was stopped was measured. In addition, the surface of the photoreceptor was subsequently irradiated with white light at an illuminance of 100 luxes, and a time (seconds) required until the  $V_d$  became one-half was determined and defined as sensitivity  $E_{1/2}$  (lux.s). Furthermore, a surface potential when the photoreceptor was irradiated with white light at an illuminance of 100 luxes was designated as a residual potential  $V_r$  (V). Moreover, with respect to the photoreceptors of Photoreceptor Examples 1-C to 6-C, since high sensitivity by long wavelength light can be expected, electrophotographic characteristics when monochromic light having a wavelength of 780 nm was used were measured at the same time. That is, the measurement was carried out in the same manner till  $V_d$ . Monochromic light (780 nm) of 1  $\mu$ W was then irradiated in place of the white light, and a time (seconds) required until the  $V_d$  became one-half was determined to obtain sensitivity  $E_{1/2}$  ( $\mu$ J/cm<sup>2</sup>). Also, a residual

potential  $V_r$  (V) when this light was irradiated on the surface of the photoreceptor for 10 seconds was measured. The results of these evaluations are shown in Table 3.

[0135]

Table 3

		White light		Monochromic light of 780 nm	
		Sensitivity (lux.s)	Residual potential (V)	Sensitivity ( $\mu\text{J}/\text{cm}^2$ )	Residual potential (V)
Photoreceptor 1-C	Example	1.2	100	0.9	80
Photoreceptor 2-C	Example	1.0	80	0.7	70
Photoreceptor 3-C	Example	0.7	80	0.7	50
Photoreceptor 4-C	Example	0.9	70	0.8	50
Photoreceptor 5-C	Example	0.8	70	0.6	40
Photoreceptor 6-C	Example	1.8	140	1.8	130
Photoreceptor 7-C	Example	1.2	90	-	-
Photoreceptor 8-C	Example	1.1	90	-	-

#### BRIEF DESCRIPTION OF THE DRAWINGS

[0136]

[Fig. 1] Fig. 1 is a conceptual cross-sectional view to show a single layer type electrophotographic photoreceptor.

[Fig. 2] Fig. 2 is a conceptual cross-sectional view to show a layered type electrophotographic photoreceptor.

[Fig. 3] Fig. 3 is an IR spectrum of a compound represented by the structural formula (I-A-3).

[Fig. 4] Fig. 4 is a  $^1\text{H-NMR}$  spectrum of a compound represented by the structural formula (I-A-3).

[Fig. 5] Fig. 5 is an IR spectrum of a compound represented by the structural

formula (I-A-5).

[Fig. 6] Fig. 6 is a  $^1\text{H-NMR}$  spectrum of a compound represented by the structural formula (I-A-5).

[Fig. 7] Fig. 7 is an IR spectrum of a compound represented by the structural formula (I-B-4).

[Fig. 8] Fig. 8 is a  $^1\text{H-NMR}$  spectrum of a compound represented by the structural formula (I-B-4).

[Fig. 9] Fig. 9 is an IR spectrum of a compound represented by the structural formula (I-B-7).

[Fig. 10] Fig. 10 is a  $^1\text{H-NMR}$  spectrum of a compound represented by the structural formula (I-B-7).

[Fig. 11] Fig. 11 is an IR spectrum of a compound represented by the structural formula (I-B-12).

[Fig. 12] Fig. 12 is a  $^1\text{H-NMR}$  spectrum of a compound represented by the structural formula (I-B-12).

[Fig. 13] Fig. 13 is an IR spectrum of a compound represented by the structural formula (I-C-1).

[Fig. 14] Fig. 14 is a  $^1\text{H-NMR}$  spectrum of a compound represented by the structural formula (I-C-1).

#### DESCRIPTION OF REFERENCE NUMERALS AND SIGNS

[0137]

- 1: Electrically conductive substrate
- 2: Photosensitive layer
- 3: Charge generation layer
- 4: Charge transport layer

- 5: Photosensitive layer (laminated)
- 6: Covering layer